

# benzene, 1,3,5-tris(bromomethyl)-

<b>Other names:</b>	1,3,5-tris(bromomethyl)benzene
<b>Inchi:</b>	InChI=1S/C9H9Br3/c10-4-7-1-8(5-11)3-9(2-7)6-12/h1-3H,4-6H2
<b>InchiKey:</b>	GHITVUOBZBZMND-UHFFFAOYSA-N
<b>Formula:</b>	C9H9Br3
<b>SMILES:</b>	BrCc1cc(CBr)cc(CBr)c1
<b>Mol. weight [g/mol]:</b>	356.88

## Physical Properties

Property code	Value	Unit	Source
gf	161.01	kJ/mol	Joback Method
hf	63.49	kJ/mol	Joback Method
hfus	28.18	kJ/mol	Joback Method
hvap	58.53	kJ/mol	Joback Method
log10ws	-5.40		Crippen Method
logp	4.371		Crippen Method
mvol	166.410	ml/mol	McGowan Method
pc	4036.37	kPa	Joback Method
rinpol	1968.00		NIST Webbook
rinpol	1968.00		NIST Webbook
tb	640.44	K	Joback Method
tc	895.18	K	Joback Method
tf	422.05	K	Joback Method
vc	0.618	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	319.60	J/molxK	640.44	Joback Method
cpg	329.80	J/molxK	682.90	Joback Method
cpg	339.21	J/molxK	725.35	Joback Method
cpg	347.89	J/molxK	767.81	Joback Method
cpg	355.95	J/molxK	810.27	Joback Method
cpg	363.44	J/molxK	852.73	Joback Method
cpg	370.46	J/molxK	895.18	Joback Method

dvisc	0.0011120	Paxs	422.05	Joback Method
dvisc	0.0007715	Paxs	458.45	Joback Method
dvisc	0.0005649	Paxs	494.85	Joback Method
dvisc	0.0004316	Paxs	531.25	Joback Method
dvisc	0.0003414	Paxs	567.64	Joback Method
dvisc	0.0002778	Paxs	604.04	Joback Method
dvisc	0.0002314	Paxs	640.44	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U402556&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U402556&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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