

# 1,2-Cyclohexanedicarboxylic acid, 4-octyl tetradecyl ester

Inchi:	InChI=1S/C30H56O4/c1-4-7-9-10-11-12-13-14-15-16-17-20-25-33-29(31)27-23-18-19-24
InchiKey:	PPQURVWPTQIGNM-UHFFFAOYSA-N
Formula:	C30H56O4
SMILES:	CCCCCCCCCCCCCOC(=O)C1CCCCC1C(=O)OC(CCC)CCCC
Mol. weight [g/mol]:	480.76

## Physical Properties

Property code	Value	Unit	Source
gf	-251.82	kJ/mol	Joback Method
hf	-1123.43	kJ/mol	Joback Method
hfus	68.41	kJ/mol	Joback Method
hvap	100.42	kJ/mol	Joback Method
log10ws	-9.63		Crippen Method
logp	8.939		Crippen Method
mcvol	437.580	ml/mol	McGowan Method
pc	683.14	kPa	Joback Method
rinpol	3235.00		NIST Webbook
rinpol	3235.00		NIST Webbook
tb	1052.82	K	Joback Method
tc	1301.97	K	Joback Method
tf	560.32	K	Joback Method
vc	1.690	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1604.69	J/molxK	1052.82	Joback Method
cpg	1625.59	J/molxK	1094.35	Joback Method
cpg	1643.99	J/molxK	1135.87	Joback Method
cpg	1659.98	J/molxK	1177.40	Joback Method
cpg	1673.65	J/molxK	1218.92	Joback Method
cpg	1685.07	J/molxK	1260.45	Joback Method
cpg	1694.33	J/molxK	1301.97	Joback Method
dvisc	0.0003411	Paxs	560.32	Joback Method

dvisc	0.0001438	Paxs	642.40	Joback Method
dvisc	0.0000738	Paxs	724.49	Joback Method
dvisc	0.0000433	Paxs	806.57	Joback Method
dvisc	0.0000281	Paxs	888.65	Joback Method
dvisc	0.0000196	Paxs	970.74	Joback Method
dvisc	0.0000145	Paxs	1052.82	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339528&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339528&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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