

# Isophthalic acid, dodecyl 4-tert-butylcyclohexyl ester

<b>Inchi:</b>	InChI=1S/C30H48O4/c1-5-6-7-8-9-10-11-12-13-14-22-33-28(31)24-16-15-17-25(23-24)2
<b>InchiKey:</b>	VAIVVLATSYGNBV-UHFFFAOYSA-N
<b>Formula:</b>	C30H48O4
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)c1cccc(C(=O)OC2CCC(C(C)(C)C)CC2)c1
<b>Mol. weight [g/mol]:</b>	472.70

## Physical Properties

Property code	Value	Unit	Source
gf	-143.76	kJ/mol	Joback Method
hf	-901.84	kJ/mol	Joback Method
hfus	58.17	kJ/mol	Joback Method
hvap	102.45	kJ/mol	Joback Method
log10ws	-9.85		Crippen Method
logp	8.526		Crippen Method
mcvol	413.820	ml/mol	McGowan Method
pc	816.79	kPa	Joback Method
rinpol	3548.00		NIST Webbook
rinpol	3548.00		NIST Webbook
tb	1081.69	K	Joback Method
tc	1325.22	K	Joback Method
tf	616.68	K	Joback Method
vc	1.577	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1494.16	J/molxK	1081.69	Joback Method
cpg	1511.08	J/molxK	1122.28	Joback Method
cpg	1526.04	J/molxK	1162.87	Joback Method
cpg	1539.15	J/molxK	1203.45	Joback Method
cpg	1550.51	J/molxK	1244.04	Joback Method
cpg	1560.25	J/molxK	1284.63	Joback Method
cpg	1568.47	J/molxK	1325.22	Joback Method
dvisc	0.0002018	Paxs	616.68	Joback Method

dvisc	0.0000983	Paxs	694.18	Joback Method
dvisc	0.0000554	Paxs	771.68	Joback Method
dvisc	0.0000346	Paxs	849.18	Joback Method
dvisc	0.0000234	Paxs	926.69	Joback Method
dvisc	0.0000168	Paxs	1004.19	Joback Method
dvisc	0.0000127	Paxs	1081.69	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U345743&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U345743&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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