

# Fumaric acid, 2-decyl nonadecyl ester

<b>Inchi:</b>	InChI=1S/C33H62O4/c1-4-6-8-10-12-13-14-15-16-17-18-19-20-21-22-24-26-30-36-32(34
<b>InchiKey:</b>	DCRUNYSYJXFGGA-ZQHSETAFSA-N
<b>Formula:</b>	C33H62O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OC(C)CCCCCCCC
<b>Mol. weight [g/mol]:</b>	522.84

## Physical Properties

Property code	Value	Unit	Source
gf	-163.08	kJ/mol	Joback Method
hf	-1102.11	kJ/mol	Joback Method
hfus	83.48	kJ/mol	Joback Method
hvap	106.93	kJ/mol	Joback Method
log10ws	-11.33		Crippen Method
logp	10.420		Crippen Method
mcvol	486.410	ml/mol	McGowan Method
pc	557.83	kPa	Joback Method
rinpol	3606.00		NIST Webbook
rinpol	3606.00		NIST Webbook
tb	1110.74	K	Joback Method
tc	1418.73	K	Joback Method
tf	585.91	K	Joback Method
vc	1.905	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1774.71	J/molxK	1110.74	Joback Method
cpg	1801.07	J/molxK	1162.07	Joback Method
cpg	1824.52	J/molxK	1213.40	Joback Method
cpg	1845.31	J/molxK	1264.73	Joback Method
cpg	1863.69	J/molxK	1316.06	Joback Method
cpg	1879.90	J/molxK	1367.40	Joback Method
cpg	1894.19	J/molxK	1418.73	Joback Method
dvisc	0.0001618	Paxs	585.91	Joback Method

dvisc	0.0000642	Paxs	673.38	Joback Method
dvisc	0.0000315	Paxs	760.85	Joback Method
dvisc	0.0000179	Paxs	848.33	Joback Method
dvisc	0.0000113	Paxs	935.80	Joback Method
dvisc	0.0000077	Paxs	1023.27	Joback Method
dvisc	0.0000056	Paxs	1110.74	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348600&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348600&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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