1,3-Dicaprin

Other names: (3-decanoyloxy-2-hydroxypropyl) decanoate

2-hydroxypropane-1,3-diyl bis(decanoate)

InChl=1S/C23H44O5/c1-3-5-7-9-11-13-15-17-22(25)27-19-21(24)20-28-23(26)18-16-14-

InchiKey: BPYWNJQNVNYQSQ-UHFFFAOYSA-N

Formula: C23H44O5

SMILES: CCCCCCCCC(=O)OCC(O)COC(=O)CCCCCCCC

Mol. weight [g/mol]: 400.59

CAS: 17598-93-5

Physical Properties

Property code	Value	Unit	Source
gf	-464.32	kJ/mol	Joback Method
hf	-1165.16	kJ/mol	Joback Method
hfus	61.46	kJ/mol	Joback Method
hvap	101.39	kJ/mol	Joback Method
log10ws	-6.55		Crippen Method
logp	5.715		Crippen Method
mcvol	355.680	ml/mol	McGowan Method
рс	956.14	kPa	Joback Method
rinpol	2774.10		NIST Webbook
rinpol	2774.10		NIST Webbook
tb	969.96	K	Joback Method
tc	1198.35	K	Joback Method
tf	539.11	K	Joback Method
VC	1.385	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1313.21	J/mol×K	1198.35	Joback Method
cpg	1290.04	J/mol×K	1122.22	Joback Method
cpg	1276.24	J/mol×K	1084.16	Joback Method
cpg	1260.89	J/mol×K	1046.09	Joback Method
cpg	1243.95	J/mol×K	1008.03	Joback Method

cpg	1225.37	J/mol×K	969.96	Joback Method	
cpg	1302.35	J/mol×K	1160.29	Joback Method	
dvisc	0.0000798	Paxs	610.92	Joback Method	
dvisc	0.0000339	Paxs	682.73	Joback Method	
dvisc	0.0000170	Paxs	754.53	Joback Method	
dvisc	0.0000096	Paxs	826.34	Joback Method	
dvisc	0.0000059	Paxs	898.15	Joback Method	
dvisc	0.0002359	Paxs	539.11	Joback Method	
dvisc	0.0000039	Paxs	969.96	Joback Method	
pvap	4.30	kPa	565.39	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	
pvap	3.60	kPa	562.37	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	
pvap	3.10	kPa	558.74	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	
pvap	2.50	kPa	554.25	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	
pvap	1.50	kPa	544.06	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	

pvap	1.10	kPa	538.32	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	
pvap	1.00	kPa	535.66	Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Pressures by the Differential Scanning Calorimetry Technique	

Sources

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Boiling Points of Short-Chain Partial Acylglycerols and Tocopherols at Low Prossure Bright Defice Differential Scanning Calorimetry Technique: McGowan Method:

https://www.doi.org/10.1021/je401080p

https://en.wikipedia.org/wiki/Joback_method

http://link.springer.com/article/10.1007/BF02311772

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C17598935&Units=SI

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg: Ideal gas heat capacity

dvisc: Dynamic viscosity

gf: Standard Gibbs free energy of formationhf: Enthalpy of formation at standard conditionshfus: Enthalpy of fusion at standard conditions

hvap: Enthalpy of vaporization at standard conditions

log10ws:Log10 of Water solubility in mol/llogp:Octanol/Water partition coefficientmcvol:McGowan's characteristic volume

pc: Critical Pressurepvap: Vapor pressure

rinpol: Non-polar retention indices

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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