

# DL-Alanine, N-methyl-N-(2-benzyloxyethoxycarbonyl)-, 2-benzyloxyethyl ester

InChI: InChI=1S/C23H29NO6/c1-19(22(25)29-15-13-27-17-20-9-5-3-6-10-20)24(2)23(26)30-16  
InChIKey: DIAMMBHSEGNBBB-UHFFFAOYSA-N

Formula: C23H29NO6

SMILES: CC(C(=O)OCCOCc1ccccc1)N(C)C(=O)OCCOCc1ccccc1

Mol. weight [g/mol]: 415.48

## Physical Properties

Property code	Value	Unit	Source
gf	-201.90	kJ/mol	Joback Method
hf	-736.78	kJ/mol	Joback Method
hfus	50.86	kJ/mol	Joback Method
hvap	96.13	kJ/mol	Joback Method
log10ws	-4.21		Crippen Method
logp	3.420		Crippen Method
mcvol	324.010	ml/mol	McGowan Method
pc	1371.74	kPa	Joback Method
rinpol	3027.00		NIST Webbook
rinpol	3027.00		NIST Webbook
tb	988.42	K	Joback Method
tc	1214.41	K	Joback Method
tf	608.06	K	Joback Method
vc	1.204	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1055.43	J/mol×K	988.42	Joback Method
cpg	1068.03	J/mol×K	1026.08	Joback Method
cpg	1078.99	J/mol×K	1063.75	Joback Method
cpg	1088.36	J/mol×K	1101.41	Joback Method
cpg	1096.17	J/mol×K	1139.08	Joback Method
cpg	1102.45	J/mol×K	1176.74	Joback Method
cpg	1107.25	J/mol×K	1214.41	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392698&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392698&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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