

# Glutaric acid, 3-methylbut-2-yl diphenylmethyl ester

Inchi:	InChI=1S/C23H28O4/c1-17(2)18(3)26-21(24)15-10-16-22(25)27-23(19-11-6-4-7-12-19)2
InchiKey:	ASVPDTLBQOYMPI-UHFFFAOYSA-N
Formula:	C23H28O4
SMILES:	CC(C)C(C)OC(=O)CCCC(=O)OC(c1ccccc1)c1ccccc1
Mol. weight [g/mol]:	368.47

## Physical Properties

Property code	Value	Unit	Source
gf	-107.56	kJ/mol	Joback Method
hf	-550.43	kJ/mol	Joback Method
hfus	38.41	kJ/mol	Joback Method
hvap	88.49	kJ/mol	Joback Method
log10ws	-5.81		Crippen Method
logp	5.077		Crippen Method
mcvol	302.290	ml/mol	McGowan Method
pc	1426.15	kPa	Joback Method
rinpol	2605.00		NIST Webbook
rinpol	2605.00		NIST Webbook
tb	930.26	K	Joback Method
tc	1157.35	K	Joback Method
tf	501.13	K	Joback Method
vc	1.137	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	959.43	J/molxK	930.26	Joback Method
cpg	1019.43	J/molxK	1119.50	Joback Method
cpg	1010.08	J/molxK	1081.65	Joback Method
cpg	999.47	J/molxK	1043.80	Joback Method
cpg	987.53	J/molxK	1005.96	Joback Method
cpg	974.20	J/molxK	968.11	Joback Method
cpg	1027.57	J/molxK	1157.35	Joback Method
dvisc	0.0000274	Paxs	930.26	Joback Method

dvisc	0.0000371	Paxs	858.74	Joback Method
dvisc	0.0000531	Paxs	787.22	Joback Method
dvisc	0.0000817	Paxs	715.70	Joback Method
dvisc	0.0001382	Paxs	644.17	Joback Method
dvisc	0.0002669	Paxs	572.65	Joback Method
dvisc	0.0006217	Paxs	501.13	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393340&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393340&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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