

# Carbonic acid, but-3-en-1-yl isoheptyl ester

<b>Inchi:</b>	InChI=1S/C11H20O3/c1-4-5-8-13-11(12)14-9-6-7-10(2)3/h4,10H,1,5-9H2,2-3H3
<b>InchiKey:</b>	UTKGOJDYDMHHCA-UHFFFAOYSA-N
<b>Formula:</b>	C11H20O3
<b>SMILES:</b>	C=CCCOC(=O)OCCCC(C)C
<b>Mol. weight [g/mol]:</b>	200.27

## Physical Properties

Property code	Value	Unit	Source
gf	-211.78	kJ/mol	Joback Method
hf	-527.24	kJ/mol	Joback Method
hfus	23.42	kJ/mol	Joback Method
hvap	50.59	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	3.152		Crippen Method
mvol	174.860	ml/mol	McGowan Method
pc	2075.54	kPa	Joback Method
rinpol	1302.00		NIST Webbook
rinpol	1302.00		NIST Webbook
tb	546.03	K	Joback Method
tc	722.74	K	Joback Method
tf	291.36	K	Joback Method
vc	0.668	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	422.86	J/molxK	546.03	Joback Method
cpg	437.39	J/molxK	575.48	Joback Method
cpg	451.36	J/molxK	604.93	Joback Method
cpg	464.75	J/molxK	634.38	Joback Method
cpg	477.59	J/molxK	663.84	Joback Method
cpg	489.86	J/molxK	693.29	Joback Method
cpg	501.56	J/molxK	722.74	Joback Method
dvisc	0.0029039	Paxs	291.36	Joback Method

dvisc	0.0013185	Paxs	333.81	Joback Method
dvisc	0.0007154	Paxs	376.25	Joback Method
dvisc	0.0004394	Paxs	418.69	Joback Method
dvisc	0.0002952	Paxs	461.14	Joback Method
dvisc	0.0002121	Paxs	503.58	Joback Method
dvisc	0.0001604	Paxs	546.03	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U383226&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U383226&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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