

# Ethylamine, 2-((p-bromo-«alpha»-methyl-«alpha»-phenylbenzyl)oxy)-N,N-dimethylethylamine

<b>Other names:</b>	2-((p-Bromo-«alpha»-methyl-«alpha»-phenylbenzyl)oxy)-N,N-dimethylethylamine Embramine Ethanamine, 2-(1-(4-bromophenyl)-1-phenylethoxy)-N,N-dimethyl- Mebrophenhydramine
<b>Inchi:</b>	InChI=1S/C18H22BrNO/c1-18(21-14-13-20(2)3,15-7-5-4-6-8-15)16-9-11-17(19)12-10-16
<b>InchiKey:</b>	URSRKSNFPUKGH-UHFFFAOYSA-N
<b>Formula:</b>	C18H22BrNO
<b>SMILES:</b>	CN(C)CCOC(C)(c1ccccc1)c1ccc(Br)cc1
<b>Mol. weight [g/mol]:</b>	348.28
<b>CAS:</b>	3565-72-8

## Physical Properties

Property code	Value	Unit	Source
gf	338.81	kJ/mol	Joback Method
hf	-0.37	kJ/mol	Joback Method
hfus	32.15	kJ/mol	Joback Method
hvap	70.47	kJ/mol	Joback Method
log10ws	-4.65		Crippen Method
logp	4.291		Crippen Method
mcvol	250.310	ml/mol	McGowan Method
pc	2010.90	kPa	Joback Method
rinpol	2185.00		NIST Webbook
rinpol	2185.00		NIST Webbook
tb	767.37	K	Joback Method
tc	1005.75	K	Joback Method
tf	474.90	K	Joback Method
vc	0.914	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	691.93	J/mol×K	767.37	Joback Method
cpg	709.00	J/mol×K	807.10	Joback Method
cpg	724.73	J/mol×K	846.83	Joback Method

cpg	739.23	J/mol×K	886.56	Joback Method
cpg	752.62	J/mol×K	926.29	Joback Method
cpg	765.03	J/mol×K	966.02	Joback Method
cpg	776.56	J/mol×K	1005.75	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3565728&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3565728&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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