

# Isophthalic acid, 2,6-dimethoxyphenyl ethyl ester

Inchi:	InChI=1S/C18H18O6/c1-4-23-17(19)12-7-5-8-13(11-12)18(20)24-16-14(21-2)9-6-10-15(
InchiKey:	KEJUOWFKTLQCGS-UHFFFAOYSA-N
Formula:	C18H18O6
SMILES:	CCOC(=O)c1cccc(C(=O)Oc2c(OC)cccc2OC)c1
Mol. weight [g/mol]:	330.33

## Physical Properties

Property code	Value	Unit	Source
gf	-381.23	kJ/mol	Joback Method
hf	-730.24	kJ/mol	Joback Method
hfus	37.24	kJ/mol	Joback Method
hvap	85.33	kJ/mol	Joback Method
log10ws	-4.46		Crippen Method
logp	3.100		Crippen Method
mcvol	243.580	ml/mol	McGowan Method
pc	1938.95	kPa	Joback Method
rinpol	2653.00		NIST Webbook
rinpol	2653.00		NIST Webbook
tb	876.96	K	Joback Method
tc	1103.73	K	Joback Method
tf	571.80	K	Joback Method
vc	0.911	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	718.64	J/molxK	876.96	Joback Method
cpg	766.39	J/molxK	1065.93	Joback Method
cpg	759.74	J/molxK	1028.14	Joback Method
cpg	751.62	J/molxK	990.34	Joback Method
cpg	742.05	J/molxK	952.55	Joback Method
cpg	731.05	J/molxK	914.75	Joback Method
cpg	771.55	J/molxK	1103.73	Joback Method
dvisc	0.0000432	Paxs	876.96	Joback Method

dvisc	0.0000529	Paxs	826.10	Joback Method
dvisc	0.0000666	Paxs	775.24	Joback Method
dvisc	0.0000866	Paxs	724.38	Joback Method
dvisc	0.0001172	Paxs	673.52	Joback Method
dvisc	0.0001666	Paxs	622.66	Joback Method
dvisc	0.0002521	Paxs	571.80	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344530&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344530&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/81-764-4/Isophthalic-acid-2-6-dimethoxyphenyl-ethyl-ester.pdf>

Generated by Cheméo on 2024-05-06 17:35:42.930198755 +0000 UTC m=+17306191.850776075.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.