

# Phenol, m-(m-phenoxyphenoxy)-

<b>Inchi:</b>	InChI=1S/C18H14O3/c19-14-6-4-9-16(12-14)21-18-11-5-10-17(13-18)20-15-7-2-1-3-8-15
<b>InchiKey:</b>	VTEWWGHLESEIEO-UHFFFAOYSA-N
<b>Formula:</b>	C18H14O3
<b>SMILES:</b>	Oc1cccc(Oc2cccc(Oc3ccccc3)c2)c1
<b>Mol. weight [g/mol]:</b>	278.30
<b>CAS:</b>	14200-84-1

## Physical Properties

Property code	Value	Unit	Source
gf	63.66	kJ/mol	Joback Method
hf	-158.48	kJ/mol	Joback Method
hfus	32.27	kJ/mol	Joback Method
hvap	80.99	kJ/mol	Joback Method
log10ws	-4.29		Crippen Method
logp	4.977		Crippen Method
mvol	210.810	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
tb	821.72	K	Joback Method
tc	1087.08	K	Joback Method
tf	540.58	K	Joback Method
vc	0.722	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	597.29	J/molxK	821.72	Joback Method
cpg	611.66	J/molxK	865.95	Joback Method
cpg	624.87	J/molxK	910.17	Joback Method
cpg	637.08	J/molxK	954.40	Joback Method
cpg	648.43	J/molxK	998.63	Joback Method
cpg	659.06	J/molxK	1042.85	Joback Method
cpg	669.12	J/molxK	1087.08	Joback Method
dvisc	0.0000892	Paxs	540.58	Joback Method
dvisc	0.0000438	Paxs	587.44	Joback Method

dvisc	0.0000239	Paxs	634.29	Joback Method
dvisc	0.0000142	Paxs	681.15	Joback Method
dvisc	0.0000090	Paxs	728.01	Joback Method
dvisc	0.0000060	Paxs	774.86	Joback Method
dvisc	0.0000042	Paxs	821.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14200841&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14200841&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/81-796-9/Phenol-m-m-phenoxyphenoxy.pdf>

Generated by Cheméo on 2025-02-19 12:34:57.28366652 +0000 UTC m=+3176713.130592148.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.