

# Fumaric acid, 2,4,4-trimethylpentyl 2-methylpent-3-yl ester

<b>Inchi:</b>	InChI=1S/C18H32O4/c1-8-15(13(2)3)22-17(20)10-9-16(19)21-12-14(4)11-18(5,6)7/h9-10
<b>InchiKey:</b>	WDSSORKNQLETJT-MDZDMXLPSA-N
<b>Formula:</b>	C18H32O4
<b>SMILES:</b>	CCC(OC(=O)C=CC(=O)OCC(C)CC(C)(C)C)C(C)C
<b>Mol. weight [g/mol]:</b>	312.44

## Physical Properties

Property code	Value	Unit	Source
gf	-291.42	kJ/mol	Joback Method
hf	-811.82	kJ/mol	Joback Method
hfus	30.17	kJ/mol	Joback Method
hvap	71.47	kJ/mol	Joback Method
log10ws	-4.32		Crippen Method
logp	4.136		Crippen Method
mcvol	275.060	ml/mol	McGowan Method
pc	1313.70	kPa	Joback Method
rinpola	1903.00		NIST Webbook
rinpola	1903.00		NIST Webbook
tb	763.43	K	Joback Method
tc	956.24	K	Joback Method
tf	389.28	K	Joback Method
vc	1.042	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	828.30	J/molxK	763.43	Joback Method
cpg	906.50	J/molxK	924.11	Joback Method
cpg	892.78	J/molxK	891.97	Joback Method
cpg	878.13	J/molxK	859.84	Joback Method
cpg	862.53	J/molxK	827.70	Joback Method
cpg	845.93	J/molxK	795.57	Joback Method
cpg	919.35	J/molxK	956.24	Joback Method
dvisc	0.0000390	Paxs	763.43	Joback Method

dvisc	0.0000558	Paxs	701.07	Joback Method
dvisc	0.0000856	Paxs	638.71	Joback Method
dvisc	0.0001441	Paxs	576.36	Joback Method
dvisc	0.0002750	Paxs	514.00	Joback Method
dvisc	0.0006277	Paxs	451.64	Joback Method
dvisc	0.0018664	Paxs	389.28	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405602&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405602&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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