

# 17-Octadecynoic acid

<b>Inchi:</b>	InChI=1S/C18H32O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18(19)20/h1H,3-17H2
<b>InchiKey:</b>	DZIILFGADWDKMF-UHFFFAOYSA-N
<b>Formula:</b>	C18H32O2
<b>SMILES:</b>	C#CCCCCCCCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	280.45
<b>CAS:</b>	34450-18-5

## Physical Properties

Property code	Value	Unit	Source
gf	58.01	kJ/mol	Joback Method
hf	-387.76	kJ/mol	Joback Method
hfus	51.04	kJ/mol	Joback Method
hvap	78.95	kJ/mol	Joback Method
log10ws	-6.25		Crippen Method
logp	5.556		Crippen Method
mcvol	263.320	ml/mol	McGowan Method
pc	1431.55	kPa	Joback Method
rinpol	2199.30		NIST Webbook
rinpol	2199.30		NIST Webbook
tb	747.41	K	Joback Method
tc	922.67	K	Joback Method
tf	450.34	K	Joback Method
vc	1.030	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	782.95	J/molxK	747.41	Joback Method
cpg	798.88	J/molxK	776.62	Joback Method
cpg	814.04	J/molxK	805.83	Joback Method
cpg	828.48	J/molxK	835.04	Joback Method
cpg	842.23	J/molxK	864.25	Joback Method
cpg	855.32	J/molxK	893.46	Joback Method
cpg	867.77	J/molxK	922.67	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57549e+01
Coeff. B	-6.33546e+03
Coeff. C	-1.37536e+02
Temperature range (K), min.	547.15
Temperature range (K), max.	744.15

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C34450185&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C34450185&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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