

# Octadecanedioic acid

<b>Other names:</b>	Hexadecanedicarboxylic acid 1,16-Hexadecanedicarboxylic acid 1,18-Octadecadiaoic acid 1,18-Octadecanedioic acid
<b>Inchi:</b>	InChI=1S/C18H34O4/c19-17(20)15-13-11-9-7-5-3-1-2-4-6-8-10-12-14-16-18(21)22/h1-16
<b>InchiKey:</b>	BNJOQKFENDDGSC-UHFFFAOYSA-N
<b>Formula:</b>	C18H34O4
<b>SMILES:</b>	O=C(O)CCCCCCCCCCCCCCCCC(=O)O
<b>Mol. weight [g/mol]:</b>	314.46
<b>CAS:</b>	871-70-5

## Physical Properties

Property code	Value	Unit	Source
gf	-430.80	kJ/mol	Joback Method
hf	-944.47	kJ/mol	Joback Method
hfus	53.75	kJ/mol	Joback Method
hvap	102.51	kJ/mol	Joback Method
log10ws	-5.56		Crippen Method
logp	5.397		Crippen Method
mcvol	279.360	ml/mol	McGowan Method
pc	1439.16	kPa	Joback Method
tb	903.34	K	Joback Method
tc	1108.58	K	Joback Method
tf	398.10 ± 1.20	K	NIST Webbook
vc	1.093	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	922.85	J/molxK	903.34	Joback Method
cpg	938.03	J/molxK	937.55	Joback Method
cpg	952.27	J/molxK	971.75	Joback Method
cpg	965.61	J/molxK	1005.96	Joback Method
cpg	978.10	J/molxK	1040.17	Joback Method

cpg	989.80	J/mol×K	1074.37	Joback Method
cpg	1000.75	J/mol×K	1108.58	Joback Method
dvisc	0.0003702	Paxs	514.12	Joback Method
dvisc	0.0001015	Paxs	578.99	Joback Method
dvisc	0.0000361	Paxs	643.86	Joback Method
dvisc	0.0000155	Paxs	708.73	Joback Method
dvisc	0.0000077	Paxs	773.60	Joback Method
dvisc	0.0000043	Paxs	838.47	Joback Method
dvisc	0.0000026	Paxs	903.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C871705&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C871705&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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