

# Diglycolic acid, heptyl 4-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C18H26O6/c1-3-4-5-6-7-12-23-17(19)13-22-14-18(20)24-16-10-8-15(21-2)9-1
<b>InchiKey:</b>	UFRLMYRZDRMTIG-UHFFFAOYSA-N
<b>Formula:</b>	C18H26O6
<b>SMILES:</b>	CCCCCCCOC(=O)COCC(=O)Oc1ccc(OC)cc1
<b>Mol. weight [g/mol]:</b>	338.40

## Physical Properties

Property code	Value	Unit	Source
gf	-474.38	kJ/mol	Joback Method
hf	-943.83	kJ/mol	Joback Method
hfus	43.98	kJ/mol	Joback Method
hvap	81.73	kJ/mol	Joback Method
log10ws	-3.62		Crippen Method
logp	3.131		Crippen Method
mcvol	267.340	ml/mol	McGowan Method
pc	1511.67	kPa	Joback Method
rinpol	3119.00		NIST Webbook
rinpol	3119.00		NIST Webbook
tb	840.32	K	Joback Method
tc	1041.15	K	Joback Method
tf	520.34	K	Joback Method
vc	1.020	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	818.63	J/molxK	840.32	Joback Method
cpg	880.63	J/molxK	1007.68	Joback Method
cpg	870.62	J/molxK	974.21	Joback Method
cpg	859.41	J/molxK	940.74	Joback Method
cpg	847.00	J/molxK	907.26	Joback Method
cpg	833.41	J/molxK	873.79	Joback Method
cpg	889.43	J/molxK	1041.15	Joback Method
dvisc	0.0000409	Paxs	840.32	Joback Method

dvisc	0.0000520	Paxs	786.99	Joback Method
dvisc	0.0000684	Paxs	733.66	Joback Method
dvisc	0.0000939	Paxs	680.33	Joback Method
dvisc	0.0001361	Paxs	627.00	Joback Method
dvisc	0.0002114	Paxs	573.67	Joback Method
dvisc	0.0003593	Paxs	520.34	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381888&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381888&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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