

# Isophthalic acid, 3,5-dimethylcyclohexyl ethyl ester

Inchi:	InChI=1S/C18H24O4/c1-4-21-17(19)14-6-5-7-15(11-14)18(20)22-16-9-12(2)8-13(3)10-16
InchiKey:	XBWDZWGQHVMWII-UHFFFAOYSA-N
Formula:	C18H24O4
SMILES:	CCOC(=O)c1cccc(C(=O)OC2CC(C)CC(C)C2)c1
Mol. weight [g/mol]:	304.38

## Physical Properties

Property code	Value	Unit	Source
gf	-255.35	kJ/mol	Joback Method
hf	-665.75	kJ/mol	Joback Method
hfus	35.58	kJ/mol	Joback Method
hvap	76.72	kJ/mol	Joback Method
log10ws	-4.83		Crippen Method
logp	3.845		Crippen Method
mvol	244.740	ml/mol	McGowan Method
pc	1737.56	kPa	Joback Method
rinpol	2319.00		NIST Webbook
rinpol	2319.00		NIST Webbook
tb	805.69	K	Joback Method
tc	1028.90	K	Joback Method
tf	474.78	K	Joback Method
vc	0.914	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	763.00	J/molxK	805.69	Joback Method
cpg	780.94	J/molxK	842.89	Joback Method
cpg	797.34	J/molxK	880.09	Joback Method
cpg	812.21	J/molxK	917.29	Joback Method
cpg	825.56	J/molxK	954.50	Joback Method
cpg	837.40	J/molxK	991.70	Joback Method
cpg	847.72	J/molxK	1028.90	Joback Method
dvisc	0.0009295	Paxs	474.78	Joback Method

dvisc	0.0005549	Paxs	529.93	Joback Method
dvisc	0.0003652	Paxs	585.08	Joback Method
dvisc	0.0002582	Paxs	640.24	Joback Method
dvisc	0.0001930	Paxs	695.39	Joback Method
dvisc	0.0001505	Paxs	750.54	Joback Method
dvisc	0.0001214	Paxs	805.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U343818&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U343818&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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