

# 9-Octadecenoic acid, 12-hydroxy-, methyl ester, [R-(Z)]-

Other names:	12-Hydroxy-9-octadecenoic acid methyl ester 9-octadecenoic acid, 12-hydroxy-, [R]-, methyl ester Flexricin P-1 Methyl ricinolate Ricinoleic acid methyl ester [R]-methyl 12-hydroxy-9-octadecenoate methyl ricinoleate
Inchi:	InChI=1S/C19H36O3/c1-3-4-5-12-15-18(20)16-13-10-8-6-7-9-11-14-17-19(21)22-2/h10,1
InchiKey:	XKGDWZQXVZSXAO-GFBZKKKVSA-N
Formula:	C19H36O3
SMILES:	CCCCCCC(O)CC=CCCCCCCCC(=O)OC
Mol. weight [g/mol]:	312.49
CAS:	141-24-2

## Physical Properties

Property code	Value	Unit	Source
cpl	696.53	J/molxK	Energies of combustion and standard molar enthalpies of formation of ricinoleic acid and methyl ricinoleate
gf	-183.86	kJ/mol	Joback Method
hf	-720.58	kJ/mol	Joback Method
hfus	48.52	kJ/mol	Joback Method
hvap	83.29	kJ/mol	Joback Method
log10ws	-5.87		Crippen Method
logp	5.168		Crippen Method
mcvol	287.580	ml/mol	McGowan Method
pc	1236.35	kPa	Joback Method
tb	806.31	K	Joback Method
tc	988.85	K	Joback Method
tf	416.79	K	Joback Method
vc	1.117	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	934.70	J/molxK	867.16	Joback Method
cpg	901.55	J/molxK	806.31	Joback Method
cpg	949.99	J/molxK	897.58	Joback Method
cpg	964.47	J/molxK	928.00	Joback Method
cpg	978.17	J/molxK	958.43	Joback Method
cpg	991.13	J/molxK	988.85	Joback Method
cpg	918.57	J/molxK	836.73	Joback Method
dvisc	0.0003895	Paxs	481.71	Joback Method
dvisc	0.0001383	Paxs	546.63	Joback Method
dvisc	0.0000611	Paxs	611.55	Joback Method
dvisc	0.0000316	Paxs	676.47	Joback Method
dvisc	0.0000184	Paxs	741.39	Joback Method
dvisc	0.0015155	Paxs	416.79	Joback Method
dvisc	0.0000116	Paxs	806.31	Joback Method
hvapt	89.30	kJ/mol	498.00	NIST Webbook
rho1	857.50	kg/m3	383.23	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rho1	865.80	kg/m3	373.26	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rho1	875.00	kg/m3	363.38	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rho1	883.30	kg/m3	353.06	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate

rhoI	904.00	kg/m3	333.25	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rhoI	910.30	kg/m3	323.05	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rhoI	916.70	kg/m3	313.33	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rhoI	921.20	kg/m3	303.17	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rhoI	928.50	kg/m3	293.16	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate
rhoI	895.80	kg/m3	343.32	Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C141242&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C141242&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Energies of combustion and standard molar enthalpies of formation of methyl ricinoleate:	<a href="https://www.doi.org/10.1016/j.jct.2012.02.006">https://www.doi.org/10.1016/j.jct.2012.02.006</a>
Measurements and Correlations of the Density, Viscosity, and Vapor Pressure for Methyl Ricinoleate:	<a href="https://www.doi.org/10.1021/acs.jced.5b00545">https://www.doi.org/10.1021/acs.jced.5b00545</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rho:</b>	Liquid Density
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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