

# Succinic acid, cyclohexylmethyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C19H34O4/c1-15(12-19(2,3)4)13-22-17(20)10-11-18(21)23-14-16-8-6-5-7-9-16
InchiKey:	BZYMIMLJHQJTOF-UHFFFAOYSA-N
Formula:	C19H34O4
SMILES:	CC(COC(=O)CCC(=O)OCC1CCCCC1)CC(C)(C)C
Mol. weight [g/mol]:	326.47

## Physical Properties

Property code	Value	Unit	Source
gf	-333.89	kJ/mol	Joback Method
hf	-884.80	kJ/mol	Joback Method
hfus	31.44	kJ/mol	Joback Method
hvap	74.94	kJ/mol	Joback Method
log10ws	-4.67		Crippen Method
logp	4.506		Crippen Method
mvol	282.590	ml/mol	McGowan Method
pc	1353.63	kPa	Joback Method
rinpol	2152.00		NIST Webbook
rinpol	2152.00		NIST Webbook
tb	802.58	K	Joback Method
tc	1005.59	K	Joback Method
tf	443.01	K	Joback Method
vc	1.063	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	907.36	J/molxK	802.58	Joback Method
cpg	926.64	J/molxK	836.41	Joback Method
cpg	944.58	J/molxK	870.25	Joback Method
cpg	961.22	J/molxK	904.08	Joback Method
cpg	976.60	J/molxK	937.92	Joback Method
cpg	990.76	J/molxK	971.75	Joback Method
cpg	1003.74	J/molxK	1005.59	Joback Method
dvisc	0.0012420	Paxs	443.01	Joback Method

dvisc	0.0005198	Paxs	502.94	Joback Method
dvisc	0.0002618	Paxs	562.87	Joback Method
dvisc	0.0001505	Paxs	622.79	Joback Method
dvisc	0.0000954	Paxs	682.72	Joback Method
dvisc	0.0000650	Paxs	742.65	Joback Method
dvisc	0.0000470	Paxs	802.58	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389545&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389545&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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