

# 2-Propenoic acid, 3-phenyl-, 1,7,7-trimethylbicyclo[2.2.1]hept-2-yl ester, endo-

Other names:

Bornyl cinnamate

Cinnamic acid, 2-bornyl ester, endo-

Bornyl cinnamate 1

Bornyl cinnamate 2

endo-1,7,7-trimethylbicyclo[2.2.1]hept-2-yl cinnamate

(1R,2S,4R)-1,7,7-Trimethylbicyclo[2.2.1]heptan-2-yl cinnamate

Inchi:

InChI=1S/C19H24O2/c1-18(2)15-11-12-19(18,3)16(13-15)21-17(20)10-9-14-7-5-4-6-8-14

InchiKey:

ACTRLDZRLKIJEH-MDZDMXLPSA-N

Formula:

C19H24O2

SMILES:

CC1(C)C2CCC1(C)C(OC(=O)C=Cc1cccc1)C2

Mol. weight [g/mol]:

284.39

CAS:

6330-67-2

## Physical Properties

Property code	Value	Unit	Source
gf	150.81	kJ/mol	Joback Method
hf	-197.30	kJ/mol	Joback Method
hfus	25.71	kJ/mol	Joback Method
hvap	66.36	kJ/mol	Joback Method
log10ws	-4.94		Crippen Method
logp	4.458		Crippen Method
mcvol	236.230	ml/mol	McGowan Method
pc	1883.80	kPa	Joback Method
rinpol	2222.70		NIST Webbook
rinpol	2222.70		NIST Webbook
tb	750.14	K	Joback Method
tc	989.00	K	Joback Method
tf	469.07	K	Joback Method
vc	0.895	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	713.15	J/molxK	750.14	Joback Method

cpg	734.31	J/mol×K	789.95	Joback Method
cpg	755.03	J/mol×K	829.76	Joback Method
cpg	775.69	J/mol×K	869.57	Joback Method
cpg	796.62	J/mol×K	909.38	Joback Method
cpg	818.19	J/mol×K	949.19	Joback Method
cpg	840.75	J/mol×K	989.00	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6330672&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6330672&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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