

# Succinic acid, cyclohexylmethyl 2-methoxy-5-methylphenyl ester

Inchi:	InChI=1S/C19H26O5/c1-14-8-9-16(22-2)17(12-14)24-19(21)11-10-18(20)23-13-15-6-4-3
InchiKey:	VJAGGLKAAYAABN-UHFFFAOYSA-N
Formula:	C19H26O5
SMILES:	COc1ccc(C)cc1OC(=O)CCC(=O)OCC1CCCCC1
Mol. weight [g/mol]:	334.41

## Physical Properties

Property code	Value	Unit	Source
gf	-346.14	kJ/mol	Joback Method
hf	-789.40	kJ/mol	Joback Method
hfus	36.83	kJ/mol	Joback Method
hvap	82.64	kJ/mol	Joback Method
log10ws	-4.66		Crippen Method
logp	3.813		Crippen Method
mcvol	264.700	ml/mol	McGowan Method
pc	1651.11	kPa	Joback Method
rinpol	2577.00		NIST Webbook
tb	865.31	K	Joback Method
tc	1086.00	K	Joback Method
tf	529.28	K	Joback Method
vc	0.991	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	842.13	J/molxK	865.31	Joback Method
cpg	858.22	J/molxK	902.09	Joback Method
cpg	872.77	J/molxK	938.87	Joback Method
cpg	885.79	J/molxK	975.66	Joback Method
cpg	897.28	J/molxK	1012.44	Joback Method
cpg	907.24	J/molxK	1049.22	Joback Method
cpg	915.68	J/molxK	1086.00	Joback Method
dvisc	0.0004535	Paxs	529.28	Joback Method
dvisc	0.0002630	Paxs	585.28	Joback Method

dvisc	0.0001678	Paxs	641.29	Joback Method
dvisc	0.0001150	Paxs	697.29	Joback Method
dvisc	0.0000834	Paxs	753.30	Joback Method
dvisc	0.0000633	Paxs	809.30	Joback Method
dvisc	0.0000497	Paxs	865.31	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390962&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390962&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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