

# Ethanone, 2-hydroxy-1,2-bis(4-methoxyphenyl)-

Other names:

p-Anisoin  
Benzoin, 4,4'-dimethoxy-  
4,4'-Dimethoxybenzoin  
p,p'-Dimethoxybenzoin  
Anisoin  
4,4'-anisoin

**Inchi:** InChI=1S/C16H16O4/c1-19-13-7-3-11(4-8-13)15(17)16(18)12-5-9-14(20-2)10-6-12/h3-10

**InchiKey:** LRRQSCPPOIUNGX-UHFFFAOYSA-N

**Formula:** C16H16O4

**SMILES:** COc1ccc(C(=O)C(O)c2ccc(OC)cc2)cc1

**Mol. weight [g/mol]:** 272.30

**CAS:** 119-52-8

## Physical Properties

Property code	Value	Unit	Source
gf	-188.78	kJ/mol	Joback Method
hf	-457.98	kJ/mol	Joback Method
hfus	29.04	kJ/mol	Joback Method
hvap	84.94	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	2.620		Crippen Method
mcvol	207.960	ml/mol	McGowan Method
pc	2517.59	kPa	Joback Method
tb	819.25	K	Joback Method
tc	1039.51	K	Joback Method
tf	383.00 ± 3.00	K	NIST Webbook
vc	0.770	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	596.90	J/mol×K	819.25	Joback Method
cpg	647.47	J/mol×K	1002.80	Joback Method
cpg	639.35	J/mol×K	966.09	Joback Method

cpg	630.25	J/molxK	929.38	Joback Method
cpg	620.16	J/molxK	892.67	Joback Method
cpg	609.05	J/molxK	855.96	Joback Method
cpg	654.63	J/molxK	1039.51	Joback Method
dvisc	0.0000155	Paxs	819.25	Joback Method
dvisc	0.0000221	Paxs	764.07	Joback Method
dvisc	0.0000333	Paxs	708.89	Joback Method
dvisc	0.0000538	Paxs	653.71	Joback Method
dvisc	0.0000950	Paxs	598.53	Joback Method
dvisc	0.0001882	Paxs	543.35	Joback Method
dvisc	0.0004352	Paxs	488.17	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119528&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119528&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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