

# Benzoic acid, 4-methyl-, (4-methylphenyl)methyl ester

Other names:

p-Toluic acid, p-methylbenzyl ester  
(4-Methylphenyl)methyl 4-methylbenzoate  
p-Methylbenzyl p-toluate  
4-Methylbenzyl 4-methylbenzoate  
p-Toluic acid, p-xylyl ester  
(4-methylphenyl)methyl p-toluate

**Inchi:** InChI=1S/C16H16O2/c1-12-3-7-14(8-4-12)11-18-16(17)15-9-5-13(2)6-10-15/h3-10H,11H

**InchiKey:** JSMHXVURHYHQGT-UHFFFAOYSA-N

**Formula:** C16H16O2

**SMILES:** Cc1ccc(COC(=O)c2ccc(C)cc2)cc1

**Mol. weight [g/mol]:** 240.30

**CAS:** 21086-87-3

## Physical Properties

Property code	Value	Unit	Source
gf	55.48	kJ/mol	Joback Method
hf	-168.25	kJ/mol	Joback Method
hfus	27.29	kJ/mol	Joback Method
hvap	66.24	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	3.660		Crippen Method
mcvol	196.220	ml/mol	McGowan Method
pc	2320.31	kPa	Joback Method
tb	705.09	K	Joback Method
tc	939.69	K	Joback Method
tf	420.12	K	Joback Method
vc	0.740	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	517.05	J/mol×K	705.09	Joback Method
cpg	532.86	J/mol×K	744.19	Joback Method
cpg	547.49	J/mol×K	783.29	Joback Method

cpg	560.99	J/molxK	822.39	Joback Method
cpg	573.40	J/molxK	861.49	Joback Method
cpg	584.75	J/molxK	900.59	Joback Method
cpg	595.10	J/molxK	939.69	Joback Method
dvisc	0.0009640	Paxs	420.12	Joback Method
dvisc	0.0005733	Paxs	467.62	Joback Method
dvisc	0.0003752	Paxs	515.11	Joback Method
dvisc	0.0002638	Paxs	562.61	Joback Method
dvisc	0.0001959	Paxs	610.10	Joback Method
dvisc	0.0001519	Paxs	657.59	Joback Method
dvisc	0.0001219	Paxs	705.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21086873&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21086873&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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