

# 2-Propanone, 1-(dimethylamino)-

<b>Other names:</b>	(Dimethylamino)acetone N,N-Dimethylaminoacetone 1-(Dimethylamino)-2-propanone 1-Dimethylaminoacetone (CH <sub>3</sub> ) <sub>2</sub> NCH <sub>2</sub> COCH <sub>3</sub>
<b>Inchi:</b>	InChI=1S/C5H11NO/c1-5(7)4-6(2)3/h4H2,1-3H3
<b>InchiKey:</b>	VFPKIWATTACVJR-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>11</sub> NO
<b>SMILES:</b>	CC(=O)CN(C)C
<b>Mol. weight [g/mol]:</b>	101.15
<b>CAS:</b>	15364-56-4

## Physical Properties

Property code	Value	Unit	Source
chl	-3315.90 ± 2.90	kJ/mol	NIST Webbook
gf	-26.92	kJ/mol	Joback Method
hf	-180.10 ± 2.90	kJ/mol	NIST Webbook
hfl	-223.70 ± 2.90	kJ/mol	NIST Webbook
hfus	13.33	kJ/mol	Joback Method
hvap	43.60 ± 0.30	kJ/mol	NIST Webbook
hvap	43.60 ± 0.25	kJ/mol	NIST Webbook
ie	7.71 ± 0.03	eV	NIST Webbook
ie	7.71 ± 0.05	eV	NIST Webbook
ie	8.65	eV	NIST Webbook
log10ws	0.23		Crippen Method
logp	0.137		Crippen Method
mcpvol	92.860	ml/mol	McGowan Method
pc	3731.66	kPa	Joback Method
rinpol	712.00		NIST Webbook
rinpol	712.00		NIST Webbook
tb	392.70	K	NIST Webbook
tc	556.89	K	Joback Method
tf	228.51	K	Joback Method
vc	0.340	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	168.27	J/mol×K	380.11	Joback Method
cpg	178.30	J/mol×K	409.57	Joback Method
cpg	187.90	J/mol×K	439.04	Joback Method
cpg	197.08	J/mol×K	468.50	Joback Method
cpg	205.86	J/mol×K	497.96	Joback Method
cpg	214.24	J/mol×K	527.43	Joback Method
cpg	222.23	J/mol×K	556.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15364564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15364564&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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