

# 4-Decenoic acid, 3-methyl-, (E)-

<b>Other names:</b>	(E)-3-Methyldec-4-enoic acid
<b>Inchi:</b>	InChI=1S/C11H20O2/c1-3-4-5-6-7-8-10(2)9-11(12)13/h7-8,10H,3-6,9H2,1-2H3,(H,12,13)
<b>InchiKey:</b>	WZZXDQPNKLAUNI-BQYQJAHWSA-N
<b>Formula:</b>	C11H20O2
<b>SMILES:</b>	CCCCC=CC(C)CC(=O)O
<b>Mol. weight [g/mol]:</b>	184.28
<b>CAS:</b>	22882-86-6

## Physical Properties

Property code	Value	Unit	Source
gf	-146.22	kJ/mol	Joback Method
hf	-423.24	kJ/mol	Joback Method
hfus	26.61	kJ/mol	Joback Method
hvap	63.08	kJ/mol	Joback Method
log10ws	-3.14		Crippen Method
logp	3.234		Crippen Method
mcvol	168.990	ml/mol	McGowan Method
pc	2358.78	kPa	Joback Method
rinpol	1388.40		NIST Webbook
rinpol	1388.40		NIST Webbook
tb	600.85	K	Joback Method
tc	775.71	K	Joback Method
tf	304.40	K	Joback Method
vc	0.650	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	432.27	J/molxK	600.85	Joback Method
cpg	445.15	J/molxK	629.99	Joback Method
cpg	457.44	J/molxK	659.14	Joback Method
cpg	469.13	J/molxK	688.28	Joback Method
cpg	480.28	J/molxK	717.43	Joback Method
cpg	490.89	J/molxK	746.57	Joback Method

cpg	500.99	J/mol×K	775.71	Joback Method
dvisc	0.0150556	Paxs	304.40	Joback Method
dvisc	0.0032736	Paxs	353.81	Joback Method
dvisc	0.0010346	Paxs	403.22	Joback Method
dvisc	0.0004204	Paxs	452.62	Joback Method
dvisc	0.0002040	Paxs	502.03	Joback Method
dvisc	0.0001127	Paxs	551.44	Joback Method
dvisc	0.0000686	Paxs	600.85	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C22882866&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C22882866&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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