

Diglycolic acid, 3,7-dimethyloctyl ethyl ester

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|-----------------------------|----------------------------------------------------------------------------------|
| Inchi: | InChI=1S/C16H30O5/c1-5-20-15(17)11-19-12-16(18)21-10-9-14(4)8-6-7-13(2)3/h13-14H |
| InchiKey: | PAVWILICCRIPFO-UHFFFAOYSA-N |
| Formula: | C16H30O5 |
| SMILES: | CCOC(=O)COCC(=O)OCCC(C)CCCC(C)C |
| Mol. weight [g/mol]: | 302.41 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|----------|----------------------|----------------|
| gf | -493.88 | kJ/mol | Joback Method |
| hf | -1005.95 | kJ/mol | Joback Method |
| hfus | 36.91 | kJ/mol | Joback Method |
| hvap | 71.16 | kJ/mol | Joback Method |
| log10ws | -2.85 | | Crippen Method |
| logp | 2.962 | | Crippen Method |
| mvol | 257.050 | ml/mol | McGowan Method |
| pc | 1410.13 | kPa | Joback Method |
| rmpol | 2402.00 | | NIST Webbook |
| rmpol | 2402.00 | | NIST Webbook |
| tb | 739.60 | K | Joback Method |
| tc | 920.56 | K | Joback Method |
| tf | 406.63 | K | Joback Method |
| vc | 0.986 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 764.62 | J/molxK | 739.60 | Joback Method |
| cpg | 839.25 | J/molxK | 890.40 | Joback Method |
| cpg | 826.12 | J/molxK | 860.24 | Joback Method |
| cpg | 812.09 | J/molxK | 830.08 | Joback Method |
| cpg | 797.17 | J/molxK | 799.92 | Joback Method |
| cpg | 781.34 | J/molxK | 769.76 | Joback Method |
| cpg | 851.49 | J/molxK | 920.56 | Joback Method |
| dvisc | 0.0000599 | Paxs | 739.60 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000810 | Paxs | 684.11 | Joback Method |
| dvisc | 0.0001156 | Paxs | 628.61 | Joback Method |
| dvisc | 0.0001766 | Paxs | 573.12 | Joback Method |
| dvisc | 0.0002955 | Paxs | 517.62 | Joback Method |
| dvisc | 0.0005594 | Paxs | 462.12 | Joback Method |
| dvisc | 0.0012606 | Paxs | 406.63 | Joback Method |

Sources

| | |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U382142&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

| | |
|-----------------|-------------------------------------------------|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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