

# Pimelic acid, ethyl 4-formylphenyl ester

<b>Inchi:</b>	InChI=1S/C16H20O5/c1-2-20-15(18)6-4-3-5-7-16(19)21-14-10-8-13(12-17)9-11-14/h8-12
<b>InchiKey:</b>	DLPXPVZDYMRSNX-UHFFFAOYSA-N
<b>Formula:</b>	C16H20O5
<b>SMILES:</b>	CCOC(=O)CCCCC(=O)Oc1ccc(C=O)cc1
<b>Mol. weight [g/mol]:</b>	292.33

## Physical Properties

Property code	Value	Unit	Source
gf	-380.74	kJ/mol	Joback Method
hf	-723.69	kJ/mol	Joback Method
hfus	38.71	kJ/mol	Joback Method
hvap	79.18	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	2.918		Crippen Method
mvol	228.990	ml/mol	McGowan Method
pc	1937.24	kPa	Joback Method
rinpol	2398.00		NIST Webbook
rinpol	2398.00		NIST Webbook
tb	798.38	K	Joback Method
tc	1003.67	K	Joback Method
tf	495.34	K	Joback Method
vc	0.888	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	664.44	J/molxK	798.38	Joback Method
cpg	721.37	J/molxK	969.45	Joback Method
cpg	711.90	J/molxK	935.24	Joback Method
cpg	701.48	J/molxK	901.02	Joback Method
cpg	690.11	J/molxK	866.81	Joback Method
cpg	677.76	J/molxK	832.59	Joback Method
cpg	729.91	J/molxK	1003.67	Joback Method
dvisc	0.0001022	Paxs	798.38	Joback Method

dvisc	0.0001287	Paxs	747.87	Joback Method
dvisc	0.0001676	Paxs	697.37	Joback Method
dvisc	0.0002275	Paxs	646.86	Joback Method
dvisc	0.0003252	Paxs	596.35	Joback Method
dvisc	0.0004966	Paxs	545.85	Joback Method
dvisc	0.0008267	Paxs	495.34	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U416642&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U416642&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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