

# Cyclopropanecarboxylic acid, trans-2-phenyl-, hexyl ester

<b>Inchi:</b>	InChI=1S/C16H22O2/c1-2-3-4-8-11-18-16(17)15-12-14(15)13-9-6-5-7-10-13/h5-7,9-10,1
<b>InchiKey:</b>	AITACQSZDPZEDG-UHFFFAOYSA-N
<b>Formula:</b>	C16H22O2
<b>SMILES:</b>	CCCCCCOC(=O)C1CC1c1ccccc1
<b>Mol. weight [g/mol]:</b>	246.34

## Physical Properties

Property code	Value	Unit	Source
gf	15.37	kJ/mol	Joback Method
hf	-329.38	kJ/mol	Joback Method
hfus	33.23	kJ/mol	Joback Method
hvap	62.25	kJ/mol	Joback Method
log10ws	-4.10		Crippen Method
logp	3.914		Crippen Method
mvol	209.120	ml/mol	McGowan Method
pc	1923.67	kPa	Joback Method
rinpol	1919.00		NIST Webbook
rinpol	1919.00		NIST Webbook
tb	670.52	K	Joback Method
tc	877.44	K	Joback Method
tf	382.36	K	Joback Method
vc	0.803	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	587.57	J/molxK	670.52	Joback Method
cpg	667.26	J/molxK	842.95	Joback Method
cpg	653.32	J/molxK	808.46	Joback Method
cpg	638.44	J/molxK	773.98	Joback Method
cpg	622.55	J/molxK	739.49	Joback Method
cpg	605.61	J/molxK	705.01	Joback Method
cpg	680.30	J/molxK	877.44	Joback Method
dvisc	0.0003824	Paxs	670.52	Joback Method

dvisc	0.0004521	Paxs	622.49	Joback Method
dvisc	0.0005497	Paxs	574.47	Joback Method
dvisc	0.0006926	Paxs	526.44	Joback Method
dvisc	0.0009141	Paxs	478.41	Joback Method
dvisc	0.0012835	Paxs	430.39	Joback Method
dvisc	0.0019626	Paxs	382.36	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405998&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405998&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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