

# p,p'-DDT

## Other names:

1,1'-(2,2,2-Trichloroethylidene)bis[4-chlorobenzene]  
1,1,1-Trichloro-2,2-bis(4-chloro fenyl)-ethaan  
1,1,1-Trichlor-2,2-bis(4-chlor-phenyl)-aethan  
1,1,1-Trichloro-2,2-bis(4,4'-dichlorodiphenyl)ethane  
1,1,1-Trichloro-2,2-bis(4-chlorophenyl)ethane (pp' DDT)  
1,1,1-Trichloro-2,2-bis(p-chlorophenyl)ethane  
1,1,1-Trichloro-2,2-di(4-chlorophenyl)ethane  
1,1,1-Tricloro-2,2-bis(4-cloro-fenil)-etano  
1,1,1-trichloro-2-2-bis(4-chlorophenyl)ethane  
1,1-Bis(4-chlorophenyl)-2,2,2-trichloroethane  
1,1-bis(p-Chlorophenyl)-2,2,2-trichloroethane  
1-chloro-4-[2,2,2-trichloro-1-(4-chlorophenyl)ethyl]benzene  
2,2,2-Trichloro-1,1-bis(4-chlorophenyl)ethane  
2,2-Bis(p-chlorophenyl)-1,1,1-trichloroethane  
4,4'-DDT  
4,4'-Dichlorodiphenyltrichloroethane  
Aavero-extra  
Agritan  
Anofex  
Arkotine  
Azotox  
Azotox M-33  
Benzene, 1,1'-(2,2,2-trichloroethylidene)bis[4-chloro-  
Benzochloryl  
Bosan supra  
Bovidermol  
Chlorophenothan  
Chlorophenothane  
Chlorophenotoxum  
Chlorphenothan  
Chlorphenotoxum  
Citox  
Clofenotan  
Clofenotane  
DDT  
DDT(p,p')  
Dedelo  
Deoval  
Detox  
Detoxan

Dibovan  
Dibovin  
Dicophane  
Dicophaner  
Didigam  
Didimac  
Dodat  
Dykol  
ENT 1,506  
ENT-1506  
Estonate  
Ethane, 1,1,1-Trichloro-2,2-bis(p-chlorophenyl)-  
Ethane, 1,1,1-trichloro-2,2-bis(4-chlorophenyl)-  
Genitox  
Gesafid  
Gesapon  
Gesarex  
Gesarol  
Geusapon  
Guesapon  
Guesarol  
Gyron  
Havero-extra  
Hildit  
Ivoran  
Ixodex  
Kopsol  
Micro DDT 75  
Mutoxan  
Mutoxin  
NCI-C00464  
Neocid  
Neocidol  
Neocidol, Solid  
OMS 16  
PEB1  
Parachlorocidum  
Pentachlorin  
Pentech  
Penticide  
Penticidum  
R50  
RCRA Waste number U061

Rukseam  
 Santobane  
 Tafidex  
 Trichlorobis(4'-Chlorophenyl)ethane  
 Trichlorobis(4-chlorophenyl)ethane  
 Zeidane  
 Zerdane  
 p,p'-Dichlorodiphenyltrichloroethane  
 p,p'-Dichlorodiphenyltrichloromethylmethane  
 p,p-DDT  
 pPzeidan  
 «alpha», «alpha»-Bis(p-chlorophenyl)-«beta», «beta», «beta»-trichlorethane  
 Â«alphaÂ», Â«alphaÂ»-Bis(p-chlorophenyl)-Â«betaÂ», Â«betaÂ», Â«betaÂ»-trichlorethan

**Inchi:** InChI=1S/C14H9Cl5/c15-11-5-1-9(2-6-11)13(14(17,18)19)10-3-7-12(16)8-4-10/h1-8,13H  
**InchiKey:** YVGGHNCTFXOJCH-UHFFFAOYSA-N  
**Formula:** C14H9Cl5  
**SMILES:** Clc1ccc(C(c2ccc(Cl)cc2)C(Cl)(Cl)Cl)cc1  
**Mol. weight [g/mol]:** 354.49  
**CAS:** 50-29-3

## Physical Properties

Property code	Value	Unit	Source
gf	213.31	kJ/mol	Joback Method
hf	25.10	kJ/mol	Joback Method
hfus	29.37	kJ/mol	Joback Method
hvap	72.88	kJ/mol	Joback Method
log10ws	-7.40		Aqueous Solubility Prediction Method
log10ws	-7.15		Estimated Solubility Method
logp	6.495		Crippen Method
mcvol	221.800	ml/mol	McGowan Method
pc	2293.71	kPa	Joback Method
rinpol	2300.00		NIST Webbook
rinpol	2300.00		NIST Webbook
rinpol	2299.00		NIST Webbook
rinpol	2270.00		NIST Webbook
rinpol	2325.20		NIST Webbook
rinpol	2330.00		NIST Webbook
rinpol	2290.00		NIST Webbook

rinpol	2289.00		NIST Webbook
rinpol	2301.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2330.00		NIST Webbook
rinpol	2290.00		NIST Webbook
rinpol	2325.20		NIST Webbook
rinpol	2263.00		NIST Webbook
rinpol	2270.00		NIST Webbook
rinpol	2277.00		NIST Webbook
rinpol	2264.00		NIST Webbook
rinpol	2270.00		NIST Webbook
rinpol	2277.00		NIST Webbook
rinpol	2264.00		NIST Webbook
rinpol	2300.00		NIST Webbook
rinpol	2270.00		NIST Webbook
rinpol	2289.00		NIST Webbook
ripol	3106.00		NIST Webbook
ripol	3103.00		NIST Webbook
ripol	3103.00		NIST Webbook
ripol	3106.00		NIST Webbook
ripol	3106.00		NIST Webbook
tb	766.52	K	Joback Method
tc	1041.34	K	Joback Method
tf	381.95	K	Aqueous Solubility Prediction Method
tf	383.17 ± 0.20	K	NIST Webbook
tf	383.00 ± 0.20	K	NIST Webbook
tf	382.25 ± 1.00	K	NIST Webbook
vc	0.832	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	495.29	J/mol×K	766.52	Joback Method
cpg	506.48	J/mol×K	812.32	Joback Method
cpg	516.50	J/mol×K	858.13	Joback Method
cpg	525.51	J/mol×K	903.93	Joback Method
cpg	533.67	J/mol×K	949.73	Joback Method
cpg	541.13	J/mol×K	995.53	Joback Method
cpg	548.06	J/mol×K	1041.34	Joback Method

dvisc	0.0001157	Paxs	715.84	Joback Method
dvisc	0.0001547	Paxs	665.16	Joback Method
dvisc	0.0002169	Paxs	614.48	Joback Method
dvisc	0.0003233	Paxs	563.80	Joback Method
dvisc	0.0005214	Paxs	513.12	Joback Method
dvisc	0.0000899	Paxs	766.52	Joback Method
dvisc	0.0009338	Paxs	462.44	Joback Method
hfust	26.28	kJ/mol	382.10	NIST Webbook
hsubt	118.00	kJ/mol	356.00	NIST Webbook
hsubt	115.00	kJ/mol	343.00	NIST Webbook
hsubt	120.20 ± 1.00	kJ/mol	293.00	NIST Webbook
hsubt	84.00	kJ/mol	338.00	NIST Webbook
hsubt	117.80	kJ/mol	303.00	NIST Webbook
hsubt	110.00	kJ/mol	323.00	NIST Webbook
hsubt	117.50	kJ/mol	343.00	NIST Webbook
hvapt	106.10 ± 1.30	kJ/mol	398.00	NIST Webbook
hvapt	93.20	kJ/mol	398.00	NIST Webbook
hvapt	338.00	kJ/mol	338.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C50293&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C50293&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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