

# Dodecane, 1,1-dimethoxy-

<b>Other names:</b>	Lauraldehyde, dimethyl acetal Dodecanal dimethyl acetal Laural dimethyl acetal n-Dodecanal dimethyl acetal 1,1-dimethoxydodecane
<b>Inchi:</b>	InChI=1S/C14H30O2/c1-4-5-6-7-8-9-10-11-12-13-14(15-2)16-3/h14H,4-13H2,1-3H3
<b>InchiKey:</b>	AJUWUYJULVYGRA-UHFFFAOYSA-N
<b>Formula:</b>	C14H30O2
<b>SMILES:</b>	CCCCCCCCCCCC(OC)OC
<b>Mol. weight [g/mol]:</b>	230.39
<b>CAS:</b>	14620-52-1

## Physical Properties

Property code	Value	Unit	Source
gf	-145.44	kJ/mol	Joback Method
hf	-602.01	kJ/mol	Joback Method
hfus	30.87	kJ/mol	Joback Method
hvap	51.19	kJ/mol	Joback Method
log10ws	-4.46		Crippen Method
logp	4.526		Crippen Method
mvol	219.860	ml/mol	McGowan Method
pc	1493.04	kPa	Joback Method
ripol	1769.00		NIST Webbook
ripol	1769.00		NIST Webbook
tb	564.12	K	Joback Method
tc	726.65	K	Joback Method
tf	277.00	K	Joback Method
vc	0.850	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	566.93	J/molxK	564.12	Joback Method
cpg	584.79	J/molxK	591.21	Joback Method

cpg	602.00	J/mol×K	618.30	Joback Method
cpg	618.57	J/mol×K	645.39	Joback Method
cpg	634.51	J/mol×K	672.48	Joback Method
cpg	649.82	J/mol×K	699.57	Joback Method
cpg	664.51	J/mol×K	726.65	Joback Method
dvisc	0.0037110	Paxs	277.00	Joback Method
dvisc	0.0013434	Paxs	324.85	Joback Method
dvisc	0.0006313	Paxs	372.71	Joback Method
dvisc	0.0003523	Paxs	420.56	Joback Method
dvisc	0.0002215	Paxs	468.41	Joback Method
dvisc	0.0001518	Paxs	516.27	Joback Method
dvisc	0.0001109	Paxs	564.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14620521&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14620521&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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