

# N,N'-di-n-Butyl-1,6-hexanediamine

<b>Other names:</b>	N,N'-di-n-Butyl hexa methylene diamine 1,6-Hexanediamine, N,N'-dibutyl- N,N'-Dibutylhexamethylenediamine Dbhmd Dibutylhexamethylenediamine N,N'-Dibutyl-1,6-hexanediamine N,N'-dibutylhexane-1,6-diamine
<b>Inchi:</b>	InChI=1S/C14H32N2/c1-3-5-11-15-13-9-7-8-10-14-16-12-6-4-2/h15-16H,3-14H2,1-2H3
<b>InchiKey:</b>	VZRUGPJUVWRHKM-UHFFFAOYSA-N
<b>Formula:</b>	C14H32N2
<b>SMILES:</b>	CCCCNCCCCCNCCCC
<b>Mol. weight [g/mol]:</b>	228.42
<b>CAS:</b>	4835-11-4

## Physical Properties

Property code	Value	Unit	Source
gf	245.78	kJ/mol	Joback Method
hf	-225.35	kJ/mol	Joback Method
hfus	42.21	kJ/mol	Joback Method
hvap	59.63	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.326		Crippen Method
mvol	228.080	ml/mol	McGowan Method
pc	1554.90	kPa	Joback Method
tb	620.06	K	Joback Method
tc	786.46	K	Joback Method
tf	352.86	K	Joback Method
vc	0.889	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	630.32	J/mol×K	620.06	Joback Method
cpg	648.07	J/mol×K	647.79	Joback Method

cpg	665.08	J/mol×K	675.53	Joback Method
cpg	681.34	J/mol×K	703.26	Joback Method
cpg	696.90	J/mol×K	731.00	Joback Method
cpg	711.76	J/mol×K	758.73	Joback Method
cpg	725.96	J/mol×K	786.46	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4835114&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4835114&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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