

# Butanoic acid, 3-methyl-, 3-phenyl-2-propenyl ester

Other names:

Isovaleric acid, cinnamyl ester

Cinnamyl isovalerate

3-Phenylallyl isovalerate

Inchi:

InChI=1S/C14H18O2/c1-12(2)11-14(15)16-10-6-9-13-7-4-3-5-8-13/h3-9,12H,10-11H2,1-

InchiKey:

FOCMOGKCPPTERB-RMKNXTFCSA-N

Formula:

C14H18O2

SMILES:

CC(C)CC(=O)OCC=Cc1ccccc1

Mol. weight [g/mol]:

218.29

CAS:

140-27-2

## Physical Properties

Property code	Value	Unit	Source
gf	23.27	kJ/mol	Joback Method
hf	-228.62	kJ/mol	Joback Method
hfus	25.52	kJ/mol	Joback Method
hvap	57.76	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	3.289		Crippen Method
mcvol	187.500	ml/mol	McGowan Method
pc	2214.53	kPa	Joback Method
ripol	1655.00		NIST Webbook
ripol	1670.00		NIST Webbook
ripol	1663.00		NIST Webbook
ripol	1670.00		NIST Webbook
ripol	1663.00		NIST Webbook
ripol	1655.00		NIST Webbook
ripol	2289.00		NIST Webbook
ripol	2289.00		NIST Webbook
ripol	2271.00		NIST Webbook
ripol	2329.00		NIST Webbook
ripol	2289.00		NIST Webbook
ripol	2271.00		NIST Webbook
tb	626.41	K	Joback Method
tc	838.56	K	Joback Method
tf	326.04	K	Joback Method
vc	0.710	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	473.76	J/molxK	626.41	Joback Method
cpg	545.17	J/molxK	803.20	Joback Method
cpg	532.72	J/molxK	767.84	Joback Method
cpg	519.40	J/molxK	732.49	Joback Method
cpg	505.16	J/molxK	697.13	Joback Method
cpg	489.96	J/molxK	661.77	Joback Method
cpg	556.80	J/molxK	838.56	Joback Method
dvisc	0.0001210	Paxs	626.41	Joback Method
dvisc	0.0001607	Paxs	576.35	Joback Method
dvisc	0.0002252	Paxs	526.29	Joback Method
dvisc	0.0003388	Paxs	476.23	Joback Method
dvisc	0.0005610	Paxs	426.16	Joback Method
dvisc	0.0010623	Paxs	376.10	Joback Method
dvisc	0.0024476	Paxs	326.04	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C140272&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C140272&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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