

# Isoamyl cinnamate

<b>Other names:</b>	Isopentyl cinnamate 2-Propenoic acid, 3-phenyl-, 3-methylbutyl ester
<b>Inchi:</b>	InChI=1S/C14H18O2/c1-12(2)10-11-16-14(15)9-8-13-6-4-3-5-7-13/h3-9,12H,10-11H2,1-14H3
<b>InchiKey:</b>	JFHCDEYLWGVZMX-CMDGGOBGSA-N
<b>Formula:</b>	C14H18O2
<b>SMILES:</b>	<chem>CC(C)CCOC(=O)C=Cc1ccccc1</chem>
<b>Mol. weight [g/mol]:</b>	218.29
<b>CAS:</b>	7779-65-9

## Physical Properties

Property code	Value	Unit	Source
gf	23.27	kJ/mol	Joback Method
hf	-228.62	kJ/mol	Joback Method
hfus	25.52	kJ/mol	Joback Method
hvap	57.76	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	3.289		Crippen Method
mcvol	187.500	ml/mol	McGowan Method
pc	2214.53	kPa	Joback Method
rinpol	1717.60		NIST Webbook
rinpol	1719.00		NIST Webbook
rinpol	1719.00		NIST Webbook
rinpol	1717.60		NIST Webbook
ripol	2355.00		NIST Webbook
ripol	2355.00		NIST Webbook
ripol	2367.00		NIST Webbook
ripol	2367.00		NIST Webbook
tb	626.41	K	Joback Method
tc	838.56	K	Joback Method
tf	326.04	K	Joback Method
vc	0.710	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	473.76	J/mol×K	626.41	Joback Method
cpg	489.96	J/mol×K	661.77	Joback Method
cpg	505.16	J/mol×K	697.13	Joback Method
cpg	519.40	J/mol×K	732.49	Joback Method
cpg	532.72	J/mol×K	767.84	Joback Method
cpg	545.17	J/mol×K	803.20	Joback Method
cpg	556.80	J/mol×K	838.56	Joback Method
dvisc	0.0024476	Paxs	326.04	Joback Method
dvisc	0.0010623	Paxs	376.10	Joback Method
dvisc	0.0005610	Paxs	426.16	Joback Method
dvisc	0.0003388	Paxs	476.23	Joback Method
dvisc	0.0002252	Paxs	526.29	Joback Method
dvisc	0.0001607	Paxs	576.35	Joback Method
dvisc	0.0001210	Paxs	626.41	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7779659&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7779659&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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