

# trans-Cinnamyl tiglate

**Other names:**

(2E)-3-Phenyl-2-propenyl (2E)-2-methyl-2-butenoate  
(E)-Cinnamyl tiglate

2-Butenoic acid, 2-methyl-, 3-phenyl-2-propenyl ester, (E,E)-

**Inchi:**

InChI=1S/C14H16O2/c1-3-12(2)14(15)16-11-7-10-13-8-5-4-6-9-13/h3-10H,11H2,1-2H3/t

**InchiKey:**

KRNURAJANZKGQN-IBIBRXRCSA-N

**Formula:**

C14H16O2

**SMILES:**

CC=C(C)C(=O)OCC=Cc1ccccc1

**Mol. weight [g/mol]:**

216.28

**CAS:**

72934-01-1

## Physical Properties

Property code	Value	Unit	Source
gf	97.38	kJ/mol	Joback Method
hf	-115.91	kJ/mol	Joback Method
hfus	27.94	kJ/mol	Joback Method
hvap	58.19	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.209		Crippen Method
mcvol	183.200	ml/mol	McGowan Method
pc	2320.31	kPa	Joback Method
ripol	2556.00		NIST Webbook
ripol	2556.00		NIST Webbook
tb	630.89	K	Joback Method
tc	851.13	K	Joback Method
tf	322.00	K	Joback Method
vc	0.697	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	452.00	J/mol×K	630.89	Joback Method
cpg	467.53	J/mol×K	667.60	Joback Method
cpg	482.04	J/mol×K	704.30	Joback Method
cpg	495.58	J/mol×K	741.01	Joback Method

cpg	508.22	J/mol×K	777.72	Joback Method
cpg	520.02	J/mol×K	814.43	Joback Method
cpg	531.05	J/mol×K	851.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72934011&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72934011&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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