

# Benzilic acid

**Other names:**

2,2-Diphenyl-2-hydroxyacetic acid  
2-Hydroxy-2,2-diphenylacetic acid  
Acide diphenylhydroxyacetique  
Benzeneacetic acid, «alpha»-hydroxy-«alpha»-phenyl-  
Benzeneacetic acid, Â«alphaÂ»-hydroxy-Â«alphaÂ»-phenyl-  
Benzylic acid  
Diphenylglycolic acid  
Diphenylhydroxyacetic acid  
Glycolic acid, diphenyl-  
Hydroxydiphenylacetic acid  
Mandelic acid, «alpha»-phenyl-  
Mandelic acid, Â«alphaÂ»-phenyl-  
NSC 2830

«alpha», «alpha»-Diphenyl-«alpha»-hydroxyacetic acid  
«alpha», «alpha»-Diphenylglycolic acid  
«alpha»-Hydroxy-2,2-diphenylacetic acid  
«alpha»-Hydroxy-«alpha»-phenylbenzeneacetic acid  
«alpha»-Hydroxydiphenylacetic acid

Â«alphaÂ», Â«alphaÂ»-Diphenyl-Â«alphaÂ»-hydroxyacetic acid  
Â«alphaÂ», Â«alphaÂ»-Diphenylglycolic acid  
Â«alphaÂ»-Hydroxy-2,2-diphenylacetic acid  
Â«alphaÂ»-Hydroxy-Â«alphaÂ»-phenylbenzeneacetic acid  
Â«alphaÂ»-Hydroxydiphenylacetic acid

**Inchi:**

InChI=1S/C14H12O3/c15-13(16)14(17,11-7-3-1-4-8-11)12-9-5-2-6-10-12/h1-10,17H,(H,1

**InchiKey:**

UKXSKSHDVLQNKG-UHFFFAOYSA-N

**Formula:**

C14H12O3

**SMILES:**

O=C(O)C(O)(c1ccccc1)c1ccccc1

**Mol. weight [g/mol]:**

228.24

**CAS:**

76-93-7

## Physical Properties

Property code	Value	Unit	Source
gf	-107.90	kJ/mol	Joback Method
hf	-285.02	kJ/mol	Joback Method
hfus	22.46	kJ/mol	Joback Method
hvap	90.12	kJ/mol	Joback Method

log10ws	-2.21		Aqueous Solubility Prediction Method
logp	2.007		Crippen Method
mcvol	173.910	ml/mol	McGowan Method
pc	3745.38	kPa	Joback Method
tb	808.08	K	Joback Method
tc	1029.71	K	Joback Method
tf	423.52	K	Aqueous Solubility Prediction Method
tf	423.00 ± 3.00	K	NIST Webbook
tf	422.00 ± 3.00	K	NIST Webbook
vc	0.636	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	490.32	J/mol×K	808.08	Joback Method
cpg	531.07	J/mol×K	992.77	Joback Method
cpg	524.11	J/mol×K	955.83	Joback Method
cpg	516.64	J/mol×K	918.89	Joback Method
cpg	508.58	J/mol×K	881.96	Joback Method
cpg	499.84	J/mol×K	845.02	Joback Method
cpg	537.61	J/mol×K	1029.71	Joback Method
dvisc	0.0000052	Paxs	808.08	Joback Method
dvisc	0.0000087	Paxs	752.46	Joback Method
dvisc	0.0000158	Paxs	696.84	Joback Method
dvisc	0.0000317	Paxs	641.23	Joback Method
dvisc	0.0000730	Paxs	585.61	Joback Method
dvisc	0.0001999	Paxs	529.99	Joback Method
dvisc	0.0006931	Paxs	474.37	Joback Method

## Sources

- Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>
- McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C76937&Units=SI>
- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>
- Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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