

# 2-(2-Chloroethoxy)-1,1'-biphenyl

<b>Inchi:</b>	InChI=1S/C14H13ClO/c15-10-11-16-14-9-5-4-8-13(14)12-6-2-1-3-7-12/h1-9H,10-11H2
<b>InchiKey:</b>	CGVSPABEGAOUOK-UHFFFAOYSA-N
<b>Formula:</b>	C14H13ClO
<b>SMILES:</b>	CICCOc1ccccc1-c1ccccc1
<b>Mol. weight [g/mol]:</b>	232.71
<b>CAS:</b>	607-04-5

## Physical Properties

Property code	Value	Unit	Source
gf	165.26	kJ/mol	Joback Method
hf	-18.66	kJ/mol	Joback Method
hfus	25.09	kJ/mol	Joback Method
hvap	58.77	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	3.971		Crippen Method
mvol	178.710	ml/mol	McGowan Method
pc	2579.34	kPa	Joback Method
tb	637.91	K	Joback Method
tc	878.20	K	Joback Method
tf	365.05	K	Joback Method
vc	0.670	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	425.04	J/molxK	637.91	Joback Method
cpg	492.07	J/molxK	838.15	Joback Method
cpg	480.81	J/molxK	798.10	Joback Method
cpg	468.52	J/molxK	758.05	Joback Method
cpg	455.17	J/molxK	718.01	Joback Method
cpg	440.69	J/molxK	677.96	Joback Method
cpg	502.37	J/molxK	878.20	Joback Method
dvisc	0.0001371	Paxs	637.91	Joback Method
dvisc	0.0001731	Paxs	592.43	Joback Method

dvisc	0.0002272	Paxs	546.96	Joback Method
dvisc	0.0003133	Paxs	501.48	Joback Method
dvisc	0.0004606	Paxs	456.00	Joback Method
dvisc	0.0007375	Paxs	410.53	Joback Method
dvisc	0.0013278	Paxs	365.05	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C607045&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C607045&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/82-558-2/2-2-Chloroethoxy-1-1-biphenyl.pdf>

Generated by Cheméo on 2024-04-28 21:01:54.318871388 +0000 UTC m=+16627363.239448704.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.