

# Pimelic acid, ethyl pentyl ester

<b>Inchi:</b>	InChI=1S/C14H26O4/c1-3-5-9-12-18-14(16)11-8-6-7-10-13(15)17-4-2/h3-12H2,1-2H3
<b>InchiKey:</b>	DVSYQMUZHLOEMY-UHFFFAOYSA-N
<b>Formula:</b>	C14H26O4
<b>SMILES:</b>	CCCCCOC(=O)CCCCC(=O)OCC
<b>Mol. weight [g/mol]:</b>	258.35

## Physical Properties

Property code	Value	Unit	Source
gf	-400.84	kJ/mol	Joback Method
hf	-821.89	kJ/mol	Joback Method
hfus	37.59	kJ/mol	Joback Method
hvap	65.07	kJ/mol	Joback Method
log10ws	-3.41		Crippen Method
logp	3.233		Crippen Method
mvol	223.000	ml/mol	McGowan Method
pc	1639.10	kPa	Joback Method
rinpol	1781.00		NIST Webbook
rinpol	1781.00		NIST Webbook
tb	672.30	K	Joback Method
tc	848.26	K	Joback Method
tf	391.86	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	622.19	J/mol×K	672.30	Joback Method
cpg	637.83	J/mol×K	701.63	Joback Method
cpg	652.75	J/mol×K	730.95	Joback Method
cpg	666.95	J/mol×K	760.28	Joback Method
cpg	680.44	J/mol×K	789.61	Joback Method
cpg	693.22	J/mol×K	818.93	Joback Method
cpg	705.29	J/mol×K	848.26	Joback Method
dvisc	0.0014859	Paxs	391.86	Joback Method

dvisc	0.0007810	Paxs	438.60	Joback Method
dvisc	0.0004646	Paxs	485.34	Joback Method
dvisc	0.0003028	Paxs	532.08	Joback Method
dvisc	0.0002115	Paxs	578.82	Joback Method
dvisc	0.0001558	Paxs	625.56	Joback Method
dvisc	0.0001198	Paxs	672.30	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393806&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393806&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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