

# 1,1,2,2-Ethanetetracarboxylic acid, tetraethyl ester

<b>Other names:</b>	Tetraethyl 1,1,2,2-ethanetetracarboxylate Tetrakis(ethoxycarbonyl)ethane 1,1,2,2-Tetracarbethoxyethane Ethane-1,1,2,2-tetracarboxylic acid, tetraethyl ester tetraethyl ethane-1,1,2,2-tetracarboxylate
<b>Inchi:</b>	InChI=1S/C14H22O8/c1-5-19-11(15)9(12(16)20-6-2)10(13(17)21-7-3)14(18)22-8-4/h9-10
<b>InchiKey:</b>	UQSBVZIXVVORQC-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O8
<b>SMILES:</b>	CCOC(=O)C(C(=O)OCC)C(C(=O)OCC)C(=O)OCC
<b>Mol. weight [g/mol]:</b>	318.32
<b>CAS:</b>	632-56-4

## Physical Properties

Property code	Value	Unit	Source
gf	-873.56	kJ/mol	Joback Method
hf	-1322.05	kJ/mol	Joback Method
hfus	36.12	kJ/mol	Joback Method
hvap	82.61	kJ/mol	Joback Method
log10ws	-0.65		Crippen Method
logp	0.471		Crippen Method
mcvol	237.880	ml/mol	McGowan Method
pc	1792.42	kPa	Joback Method
tb	824.00	K	Joback Method
tc	1020.93	K	Joback Method
tf	506.18	K	Joback Method
vc	0.903	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	716.95	J/mol×K	824.00	Joback Method
cpg	769.98	J/mol×K	988.11	Joback Method
cpg	761.55	J/mol×K	955.29	Joback Method
cpg	752.01	J/mol×K	922.47	Joback Method

cpg	741.38	J/molxK	889.64	Joback Method
cpg	729.69	J/molxK	856.82	Joback Method
cpg	777.28	J/molxK	1020.93	Joback Method
dvisc	0.0000541	Paxs	824.00	Joback Method
dvisc	0.0000705	Paxs	771.03	Joback Method
dvisc	0.0000955	Paxs	718.06	Joback Method
dvisc	0.0001359	Paxs	665.09	Joback Method
dvisc	0.0002055	Paxs	612.12	Joback Method
dvisc	0.0003360	Paxs	559.15	Joback Method
dvisc	0.0006090	Paxs	506.18	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C632564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C632564&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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