

# Isobutyl 8-methylnonanoate

<b>Inchi:</b>	InChI=1S/C14H28O2/c1-12(2)9-7-5-6-8-10-14(15)16-11-13(3)4/h12-13H,5-11H2,1-4H3
<b>InchiKey:</b>	DTJLMFONYDSQIM-UHFFFAOYSA-N
<b>Formula:</b>	C14H28O2
<b>SMILES:</b>	CC(C)CCCCCCC(=O)OCC(C)C
<b>Mol. weight [g/mol]:</b>	228.37
<b>CAS:</b>	1215127-94-8

## Physical Properties

Property code	Value	Unit	Source
gf	-171.80	kJ/mol	Joback Method
hf	-587.65	kJ/mol	Joback Method
hfus	27.76	kJ/mol	Joback Method
hvap	55.14	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	4.182		Crippen Method
mcvol	215.560	ml/mol	McGowan Method
pc	1600.00	kPa	Joback Method
rinpol	1511.40		NIST Webbook
rinpol	1511.40		NIST Webbook
tb	595.13	K	Joback Method
tc	768.62	K	Joback Method
tf	289.70	K	Joback Method
vc	0.832	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	567.68	J/molxK	595.13	Joback Method
cpg	648.02	J/molxK	739.70	Joback Method
cpg	633.39	J/molxK	710.79	Joback Method
cpg	618.05	J/molxK	681.87	Joback Method
cpg	602.00	J/molxK	652.96	Joback Method
cpg	585.21	J/molxK	624.04	Joback Method
cpg	661.96	J/molxK	768.62	Joback Method

dvisc	0.0001338	Paxs	595.13	Joback Method
dvisc	0.0001853	Paxs	544.23	Joback Method
dvisc	0.0002745	Paxs	493.32	Joback Method
dvisc	0.0004451	Paxs	442.42	Joback Method
dvisc	0.0008184	Paxs	391.51	Joback Method
dvisc	0.0018054	Paxs	340.61	Joback Method
dvisc	0.0052596	Paxs	289.70	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1215127948&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1215127948&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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