

# Diglycolic acid, isobutyl 2-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C15H20O5/c1-11(2)8-19-14(16)9-18-10-15(17)20-13-7-5-4-6-12(13)3/h4-7,11H
<b>InchiKey:</b>	PURCQNJFMDBSHL-UHFFFAOYSA-N
<b>Formula:</b>	C15H20O5
<b>SMILES:</b>	<chem>Cc1ccccc1OC(=O)COCC(=O)OCC(C)C</chem>
<b>Mol. weight [g/mol]:</b>	280.32

## Physical Properties

Property code	Value	Unit	Source
gf	-397.08	kJ/mol	Joback Method
hf	-754.97	kJ/mol	Joback Method
hfus	31.50	kJ/mol	Joback Method
hvap	72.26	kJ/mol	Joback Method
log10ws	-2.48		Crippen Method
logp	2.116		Crippen Method
mvol	219.200	ml/mol	McGowan Method
pc	1959.60	kPa	Joback Method
rinpol	2552.00		NIST Webbook
rinpol	2552.00		NIST Webbook
tb	748.82	K	Joback Method
tc	954.33	K	Joback Method
tf	449.30	K	Joback Method
vc	0.828	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	621.47	J/molxK	748.82	Joback Method
cpg	636.13	J/molxK	783.07	Joback Method
cpg	649.79	J/molxK	817.32	Joback Method
cpg	662.46	J/molxK	851.57	Joback Method
cpg	674.12	J/molxK	885.82	Joback Method
cpg	684.78	J/molxK	920.08	Joback Method
cpg	694.42	J/molxK	954.33	Joback Method
dvisc	0.0007579	Paxs	449.30	Joback Method

dvisc	0.0004248	Paxs	499.22	Joback Method
dvisc	0.0002645	Paxs	549.14	Joback Method
dvisc	0.0001782	Paxs	599.06	Joback Method
dvisc	0.0001276	Paxs	648.98	Joback Method
dvisc	0.0000958	Paxs	698.90	Joback Method
dvisc	0.0000748	Paxs	748.82	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382005&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382005&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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