

# 2(3H)-Benzofuranone, 6-ethenylhexahydro-3,6-dimethyl-7-(1-methylethoxy) Other names: 2(3H)-Benzofuranone, hexahydro-7-isopropenyl-3,6-dimethyl-6-vinyl-, (all-5R)- [3S-(3«alpha»,3a«alpha»,6«alpha»,7«beta»,7a«beta»)]-Saussurea lactone

**Inchi:** InChI=1S/C15H22O2/c1-6-15(5)8-7-11-10(4)14(16)17-13(11)12(15)9(2)3/h6,10-13H,1-2,  
**InchiKey:** LMNJALUUIMXUQQ-UHFFFAOYSA-N  
**Formula:** C15H22O2  
**SMILES:** C=CC1(C)CCC2C(C)C(=O)OC2C1C(=C)C  
**Mol. weight [g/mol]:** 234.33  
**CAS:** 23527-07-3

## Physical Properties

Property code	Value	Unit	Source
gf	90.42	kJ/mol	Joback Method
hf	-300.22	kJ/mol	Joback Method
hfus	25.11	kJ/mol	Joback Method
hvap	54.75	kJ/mol	Joback Method
log10ws	-3.61		Crippen Method
logp	3.342		Crippen Method
mcvol	199.330	ml/mol	McGowan Method
pc	1954.41	kPa	Joback Method
rinpol	1806.30		NIST Webbook
rinpol	1806.30		NIST Webbook
tb	643.13	K	Joback Method
tc	872.16	K	Joback Method
tf	372.62	K	Joback Method
vc	0.751	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	575.05	J/molxK	643.13	Joback Method
cpg	597.18	J/molxK	681.30	Joback Method
cpg	618.07	J/molxK	719.47	Joback Method
cpg	637.86	J/molxK	757.65	Joback Method
cpg	656.69	J/molxK	795.82	Joback Method

cpg	674.69	J/mol×K	833.99	Joback Method
cpg	692.00	J/mol×K	872.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C23527073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C23527073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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