

# 9-Hydroxy-decanoic acid, methyl ester

<b>Other names:</b>	Methyl 9-hydroxydecanoate
<b>Inchi:</b>	InChI=1S/C11H22O3/c1-10(12)8-6-4-3-5-7-9-11(13)14-2/h10,12H,3-9H2,1-2H3
<b>InchiKey:</b>	RXJKTNNEUMHOQZ-UHFFFAOYSA-N
<b>Formula:</b>	C11H22O3
<b>SMILES:</b>	COC(=O)CCCCCCCC(C)O
<b>Mol. weight [g/mol]:</b>	202.29

## Physical Properties

Property code	Value	Unit	Source
gf	-331.44	kJ/mol	Joback Method
hf	-672.68	kJ/mol	Joback Method
hfus	27.60	kJ/mol	Joback Method
hvap	65.53	kJ/mol	Joback Method
log10ws	-2.66		Crippen Method
logp	2.271		Crippen Method
mcvol	179.160	ml/mol	McGowan Method
pc	2208.29	kPa	Joback Method
rinpol	1494.00		NIST Webbook
rinpol	1494.00		NIST Webbook
ripol	2083.00		NIST Webbook
ripol	2083.00		NIST Webbook
tb	619.11	K	Joback Method
tc	788.09	K	Joback Method
tf	331.71	K	Joback Method
vc	0.689	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	479.00	J/molxK	619.11	Joback Method
cpg	492.23	J/molxK	647.27	Joback Method
cpg	504.91	J/molxK	675.44	Joback Method
cpg	517.04	J/molxK	703.60	Joback Method
cpg	528.63	J/molxK	731.76	Joback Method

cpg	539.68	J/mol×K	759.93	Joback Method
cpg	550.21	J/mol×K	788.09	Joback Method
dvisc	0.0077301	Paxs	331.71	Joback Method
dvisc	0.0020635	Paxs	379.61	Joback Method
dvisc	0.0007406	Paxs	427.51	Joback Method
dvisc	0.0003267	Paxs	475.41	Joback Method
dvisc	0.0001675	Paxs	523.31	Joback Method
dvisc	0.0000960	Paxs	571.21	Joback Method
dvisc	0.0000600	Paxs	619.11	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U194643&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U194643&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/82-790-4/9-Hydroxy-decanoic-acid-methyl-ester.pdf>

Generated by Cheméo on 2024-04-29 08:59:14.310122622 +0000 UTC m=+16670403.230699937.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.