

# 1,3-Dioxane, 2-ethyl-4-pentyl, 2R,4R

<b>Inchi:</b>	InChI=1S/C11H22O2/c1-3-5-6-7-10-8-9-12-11(4-2)13-10/h10-11H,3-9H2,1-2H3/t10-,11-/
<b>InchiKey:</b>	UCGYQTHBPZXKHC-QWRGUYRKSA-N
<b>Formula:</b>	C11H22O2
<b>SMILES:</b>	CCCCC1CCOC(CC)O1
<b>Mol. weight [g/mol]:</b>	186.29

## Physical Properties

Property code	Value	Unit	Source
gf	-113.76	kJ/mol	Joback Method
hf	-500.39	kJ/mol	Joback Method
hfus	33.11	kJ/mol	Joback Method
hvap	49.22	kJ/mol	Joback Method
log10ws	-3.22		Crippen Method
logp	3.108		Crippen Method
mvol	166.730	ml/mol	McGowan Method
pc	2224.99	kPa	Joback Method
ripol	1560.00		NIST Webbook
ripol	1560.00		NIST Webbook
tb	519.86	K	Joback Method
tc	712.81	K	Joback Method
tf	270.01	K	Joback Method
vc	0.625	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	416.76	J/molxK	519.86	Joback Method
cpg	503.96	J/molxK	680.65	Joback Method
cpg	488.24	J/molxK	648.49	Joback Method
cpg	471.69	J/molxK	616.34	Joback Method
cpg	454.26	J/molxK	584.18	Joback Method
cpg	435.96	J/molxK	552.02	Joback Method
cpg	518.84	J/molxK	712.81	Joback Method
dvisc	0.0002794	Paxs	519.86	Joback Method

dvisc	0.0003700	Paxs	478.22	Joback Method
dvisc	0.0005170	Paxs	436.58	Joback Method
dvisc	0.0007750	Paxs	394.94	Joback Method
dvisc	0.0012782	Paxs	353.29	Joback Method
dvisc	0.0024098	Paxs	311.65	Joback Method
dvisc	0.0055245	Paxs	270.01	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R191845&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R191845&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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