

# 1,2-Cyclohexanedicarboxylic acid, hexyl propyl ester

Inchi:	InChI=1S/C17H30O4/c1-3-5-6-9-13-21-17(19)15-11-8-7-10-14(15)16(18)20-12-4-2/h14-1
InchiKey:	BCPFJZXBQIQYNY-UHFFFAOYSA-N
Formula:	C17H30O4
SMILES:	CCCCCOC(=O)C1CCCCC1C(=O)OCCC
Mol. weight [g/mol]:	298.42

## Physical Properties

Property code	Value	Unit	Source
gf	-358.84	kJ/mol	Joback Method
hf	-849.83	kJ/mol	Joback Method
hfus	38.27	kJ/mol	Joback Method
hvap	71.87	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.870		Crippen Method
mvol	254.410	ml/mol	McGowan Method
pc	1495.35	kPa	Joback Method
rinpol	2025.00		NIST Webbook
rinpol	2025.00		NIST Webbook
tb	755.82	K	Joback Method
tc	950.83	K	Joback Method
tf	428.81	K	Joback Method
vc	0.968	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	789.64	J/molxK	755.82	Joback Method
cpg	872.45	J/molxK	918.33	Joback Method
cpg	858.22	J/molxK	885.83	Joback Method
cpg	842.83	J/molxK	853.33	Joback Method
cpg	826.28	J/molxK	820.82	Joback Method
cpg	808.55	J/molxK	788.32	Joback Method
cpg	885.53	J/molxK	950.83	Joback Method
dvisc	0.0001055	Paxs	755.82	Joback Method

dvisc	0.0001368	Paxs	701.32	Joback Method
dvisc	0.0001852	Paxs	646.82	Joback Method
dvisc	0.0002653	Paxs	592.31	Joback Method
dvisc	0.0004086	Paxs	537.81	Joback Method
dvisc	0.0006936	Paxs	483.31	Joback Method
dvisc	0.0013472	Paxs	428.81	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U339407&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U339407&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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