

# Hexanoic acid, 3,5,5-trimethyl-, oct-3-en-2-yl ester

Inchi:	InChI=1S/C17H32O2/c1-7-8-9-10-11-15(3)19-16(18)12-14(2)13-17(4,5)6/h10-11,14-15H
InchiKey:	NSCGWDUQUAUOFFR-ZHACJKMWSA-N
Formula:	C17H32O2
SMILES:	CCCCC=CC(C)OC(=O)CC(C)CC(C)(C)C
Mol. weight [g/mol]:	268.43

## Physical Properties

Property code	Value	Unit	Source
gf	-63.48	kJ/mol	Joback Method
hf	-541.10	kJ/mol	Joback Method
hfus	28.31	kJ/mol	Joback Method
hvap	60.48	kJ/mol	Joback Method
log10ws	-5.28		Crippen Method
logp	5.127		Crippen Method
mvol	253.530	ml/mol	McGowan Method
pc	1354.63	kPa	Joback Method
rinpol	1637.00		NIST Webbook
rinpol	1637.00		NIST Webbook
tb	664.70	K	Joback Method
tc	848.69	K	Joback Method
tf	320.85	K	Joback Method
vc	0.969	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	714.08	J/molxK	664.70	Joback Method
cpg	799.88	J/molxK	818.02	Joback Method
cpg	784.51	J/molxK	787.36	Joback Method
cpg	768.29	J/molxK	756.69	Joback Method
cpg	751.17	J/molxK	726.03	Joback Method
cpg	733.12	J/molxK	695.36	Joback Method
cpg	814.44	J/molxK	848.69	Joback Method
dvisc	0.0000685	Paxs	664.70	Joback Method

dvisc	0.0000987	Paxs	607.39	Joback Method
dvisc	0.0001535	Paxs	550.08	Joback Method
dvisc	0.0002644	Paxs	492.77	Joback Method
dvisc	0.0005254	Paxs	435.47	Joback Method
dvisc	0.0012858	Paxs	378.16	Joback Method
dvisc	0.0043326	Paxs	320.85	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406929&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406929&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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