

# Citronellyl heptanoate

<b>Inchi:</b>	InChI=1S/C17H32O2/c1-5-6-7-8-12-17(18)19-14-13-16(4)11-9-10-15(2)3/h10,16H,5-9,11
<b>InchiKey:</b>	JZTNHMKNNVAGPD-UHFFFAOYSA-N
<b>Formula:</b>	C17H32O2
<b>SMILES:</b>	CCCCCCC(=O)OCCC(C)CC=C(C)C
<b>Mol. weight [g/mol]:</b>	268.43
<b>CAS:</b>	72934-17-9

## Physical Properties

Property code	Value	Unit	Source
gf	-72.43	kJ/mol	Joback Method
hf	-536.86	kJ/mol	Joback Method
hfus	37.94	kJ/mol	Joback Method
hvap	62.24	kJ/mol	Joback Method
log10ws	-5.41		Crippen Method
logp	5.273		Crippen Method
mcvol	253.530	ml/mol	McGowan Method
pc	1332.96	kPa	Joback Method
rinpol	1819.00		NIST Webbook
rinpol	1796.00		NIST Webbook
rinpol	1819.00		NIST Webbook
rinpol	1820.30		NIST Webbook
rinpol	1796.00		NIST Webbook
rinpol	1820.30		NIST Webbook
tb	668.25	K	Joback Method
tc	844.66	K	Joback Method
tf	319.47	K	Joback Method
vc	0.987	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	711.05	J/mol×K	668.25	Joback Method
cpg	729.41	J/mol×K	697.65	Joback Method
cpg	746.91	J/mol×K	727.05	Joback Method

cpg	763.59	J/mol×K	756.46	Joback Method
cpg	779.47	J/mol×K	785.86	Joback Method
cpg	794.58	J/mol×K	815.26	Joback Method
cpg	808.95	J/mol×K	844.66	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72934179&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72934179&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/82-915-5/Citronellyl-heptanoate.pdf>

Generated by Cheméo on 2024-04-28 12:02:14.065387621 +0000 UTC m=+16594982.985964931.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.