

# 5-Phenylvaleric acid, 4-methyl-2-pentyl ester

<b>Inchi:</b>	InChI=1S/C17H26O2/c1-14(2)13-15(3)19-17(18)12-8-7-11-16-9-5-4-6-10-16/h4-6,9-10,1
<b>InchiKey:</b>	DVBANFVLFAZSGD-UHFFFAOYSA-N
<b>Formula:</b>	C17H26O2
<b>SMILES:</b>	CC(C)CC(C)OC(=O)CCCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	262.39

## Physical Properties

Property code	Value	Unit	Source
gf	-34.13	kJ/mol	Joback Method
hf	-413.04	kJ/mol	Joback Method
hfus	29.57	kJ/mol	Joback Method
hvap	64.09	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	4.377		Crippen Method
mcvol	234.070	ml/mol	McGowan Method
pc	1656.49	kPa	Joback Method
rinpol	1884.00		NIST Webbook
rinpol	1884.00		NIST Webbook
tb	690.45	K	Joback Method
tc	890.12	K	Joback Method
tf	349.93	K	Joback Method
vc	0.891	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.80	J/molxK	690.45	Joback Method
cpg	673.91	J/molxK	723.73	Joback Method
cpg	690.97	J/molxK	757.01	Joback Method
cpg	706.99	J/molxK	790.29	Joback Method
cpg	722.02	J/molxK	823.57	Joback Method
cpg	736.09	J/molxK	856.85	Joback Method
cpg	749.23	J/molxK	890.12	Joback Method
dvisc	0.0026892	Paxs	349.93	Joback Method

dvisc	0.0010374	Paxs	406.68	Joback Method
dvisc	0.0005053	Paxs	463.44	Joback Method
dvisc	0.0002880	Paxs	520.19	Joback Method
dvisc	0.0001833	Paxs	576.94	Joback Method
dvisc	0.0001265	Paxs	633.70	Joback Method
dvisc	0.0000928	Paxs	690.45	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406074&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406074&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/82-950-6/5-Phenylvaleric-acid-4-methyl-2-pentyl-ester.pdf>

Generated by Cheméo on 2024-04-20 13:19:36.74787898 +0000 UTC m=+15908425.668456296.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.