

# Ibuprofen, butyl ester

<b>Inchi:</b>	InChI=1S/C17H26O2/c1-5-6-11-19-17(18)14(4)16-9-7-15(8-10-16)12-13(2)3/h7-10,13-14
<b>InchiKey:</b>	JMYISEMZJYEMCF-UHFFFAOYSA-N
<b>Formula:</b>	C17H26O2
<b>SMILES:</b>	CCCCOC(=O)C(C)c1ccc(CC(C)C)cc1
<b>Mol. weight [g/mol]:</b>	262.39

## Physical Properties

Property code	Value	Unit	Source
gf	-43.76	kJ/mol	Joback Method
hf	-424.51	kJ/mol	Joback Method
hfus	29.18	kJ/mol	Joback Method
hvap	64.75	kJ/mol	Joback Method
log10ws	-4.59		Crippen Method
logp	4.332		Crippen Method
mcvol	234.070	ml/mol	McGowan Method
pc	1636.45	kPa	Joback Method
rinpol	1452.00		NIST Webbook
rinpol	1452.00		NIST Webbook
tb	695.43	K	Joback Method
tc	895.94	K	Joback Method
tf	362.45	K	Joback Method
vc	0.891	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.56	J/mol×K	695.43	Joback Method
cpg	735.28	J/mol×K	862.52	Joback Method
cpg	721.30	J/mol×K	829.10	Joback Method
cpg	706.37	J/mol×K	795.68	Joback Method
cpg	690.45	J/mol×K	762.27	Joback Method
cpg	673.53	J/mol×K	728.85	Joback Method
cpg	748.34	J/mol×K	895.94	Joback Method
dvisc	0.0000924	Paxs	695.43	Joback Method

dvisc	0.0001237	Paxs	639.93	Joback Method
dvisc	0.0001749	Paxs	584.44	Joback Method
dvisc	0.0002662	Paxs	528.94	Joback Method
dvisc	0.0004468	Paxs	473.44	Joback Method
dvisc	0.0008605	Paxs	417.95	Joback Method
dvisc	0.0020260	Paxs	362.45	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389604&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389604&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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