

# Butyric acid, 2-phenyl-, hept-2-yl ester

<b>Inchi:</b>	InChI=1S/C17H26O2/c1-4-6-8-11-14(3)19-17(18)16(5-2)15-12-9-7-10-13-15/h7,9-10,12-
<b>InchiKey:</b>	HXWXGZBJIIXDSZ-UHFFFAOYSA-N
<b>Formula:</b>	C17H26O2
<b>SMILES:</b>	CCCCC(C)OC(=O)C(CC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	262.39

## Physical Properties

Property code	Value	Unit	Source
gf	-34.13	kJ/mol	Joback Method
hf	-413.04	kJ/mol	Joback Method
hfus	29.57	kJ/mol	Joback Method
hvap	64.09	kJ/mol	Joback Method
log10ws	-4.98		Crippen Method
logp	4.692		Crippen Method
mcvol	234.070	ml/mol	McGowan Method
pc	1656.49	kPa	Joback Method
rinpol	1716.00		NIST Webbook
rinpol	1716.00		NIST Webbook
tb	690.45	K	Joback Method
tc	890.12	K	Joback Method
tf	349.93	K	Joback Method
vc	0.891	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.80	J/mol×K	690.45	Joback Method
cpg	673.91	J/mol×K	723.73	Joback Method
cpg	690.97	J/mol×K	757.01	Joback Method
cpg	706.99	J/mol×K	790.29	Joback Method
cpg	722.02	J/mol×K	823.57	Joback Method
cpg	736.09	J/mol×K	856.85	Joback Method
cpg	749.23	J/mol×K	890.12	Joback Method
dvisc	0.0026892	Paxs	349.93	Joback Method

dvisc	0.0010374	Paxs	406.68	Joback Method
dvisc	0.0005053	Paxs	463.44	Joback Method
dvisc	0.0002880	Paxs	520.19	Joback Method
dvisc	0.0001833	Paxs	576.94	Joback Method
dvisc	0.0001265	Paxs	633.70	Joback Method
dvisc	0.0000928	Paxs	690.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406859&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406859&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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