

# 8,9,10-trinorborn-5-ene-2-spiro-1'-(2'-isovaleroxy)

Inchi:	InChI=1S/C17H26O2/c1-12(2)9-16(18)19-15-5-3-4-8-17(15)11-13-6-7-14(17)10-13/h6-7,
InchiKey:	LOWYXDMTFRLVBY-UHFFFAOYSA-N
Formula:	C17H26O2
SMILES:	CC(C)CC(=O)OC1CCCCC12CC1C=CC2C1
Mol. weight [g/mol]:	262.39

## Physical Properties

Property code	Value	Unit	Source
gf	18.61	kJ/mol	Joback Method
hf	-391.69	kJ/mol	Joback Method
hfus	23.15	kJ/mol	Joback Method
hvap	61.29	kJ/mol	Joback Method
log10ws	-4.48		Crippen Method
logp	4.101		Crippen Method
mcvol	220.950	ml/mol	McGowan Method
pc	1918.62	kPa	Joback Method
ripol	1798.90		NIST Webbook
ripol	1798.90		NIST Webbook
ripol	1806.30		NIST Webbook
ripol	2211.20		NIST Webbook
ripol	2224.00		NIST Webbook
ripol	2211.20		NIST Webbook
tb	691.97	K	Joback Method
tc	915.55	K	Joback Method
tf	402.19	K	Joback Method
vc	0.836	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	674.50	J/molxK	691.97	Joback Method
cpg	696.40	J/molxK	729.23	Joback Method
cpg	717.11	J/molxK	766.50	Joback Method
cpg	736.82	J/molxK	803.76	Joback Method

cpg	755.72	J/mol×K	841.02	Joback Method
cpg	773.99	J/mol×K	878.29	Joback Method
cpg	791.84	J/mol×K	915.55	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R327801&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R327801&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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