

# Diglycolic acid, 2,5-dimethylphenyl pentyl ester

Inchi:	InChI=1S/C17H24O5/c1-4-5-6-9-21-16(18)11-20-12-17(19)22-15-10-13(2)7-8-14(15)3/h
InchiKey:	XSJISICCBROPGT-UHFFFAOYSA-N
Formula:	C17H24O5
SMILES:	CCCCCOC(=O)COCC(=O)Oc1cc(C)ccc1C
Mol. weight [g/mol]:	308.37

## Physical Properties

Property code	Value	Unit	Source
gf	-387.43	kJ/mol	Joback Method
hf	-802.44	kJ/mol	Joback Method
hfus	39.81	kJ/mol	Joback Method
hvap	77.76	kJ/mol	Joback Method
log10ws	-3.62		Crippen Method
logp	2.959		Crippen Method
mvol	247.380	ml/mol	McGowan Method
pc	1633.81	kPa	Joback Method
rmpol	2702.00		NIST Webbook
rmpol	2702.00		NIST Webbook
tb	800.00	K	Joback Method
tc	1001.07	K	Joback Method
tf	499.36	K	Joback Method
vc	0.946	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	731.87	J/molxK	800.00	Joback Method
cpg	746.79	J/molxK	833.51	Joback Method
cpg	760.66	J/molxK	867.02	Joback Method
cpg	773.48	J/molxK	900.54	Joback Method
cpg	785.25	J/molxK	934.05	Joback Method
cpg	795.95	J/molxK	967.56	Joback Method
cpg	805.60	J/molxK	1001.07	Joback Method
dvisc	0.0004802	Paxs	499.36	Joback Method

dvisc	0.0002932	Paxs	549.47	Joback Method
dvisc	0.0001945	Paxs	599.57	Joback Method
dvisc	0.0001374	Paxs	649.68	Joback Method
dvisc	0.0001020	Paxs	699.79	Joback Method
dvisc	0.0000788	Paxs	749.89	Joback Method
dvisc	0.0000629	Paxs	800.00	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382706&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382706&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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